The National Academy of Sciences, India (Registered under Act XXI of 1860) Founded 1930

COUNCIL FOR 2001

President

1. Prof. S.K. Joshi, D.Phil., D.Sc. (h.c.), F.N.A.Sc., F.A.Sc., F.N.A., F.T.W.A.S., New Delhi.

Two Past Presidents (including the Immediate Past President)

- 2. Prof. M.G.K. Menon, Ph.D. (Bristol), D.Sc. (h.c.), F.N.A.Sc., F.A.Sc., F.N.A., F.T.W.A.S., F.R.S., Delhi.
- Dr. V.P. Sharma, D.Phil., D.Sc., F.A.M.S., F.E.S.I., F.I.S.C.D., F.N.A.Sc., F.A.Sc., F.N.A., F.R.A.S., New Delhi.

Vice-Presidents

- 4. Prof. B.N. Dhawan, M.D., F.A.M.S., F.N.A.Sc., F.N.A., F.T.W.A.S., Lucknow.
- 5. Dr. Amit Ghosh, Ph.D., F.N.A.Sc., Chandigarh.

Treasurer

6. Prof. M.P. Tandon, D.Phil., F.N.A.Sc., F.I.P.S., Allahabad.

Foreign Secretary

7. Prof. Kasturi Datta, Ph.D., F.N.A.Sc., F.A.Sc., F.N.A., New Delhi.

General Secretaries

- 8. Prof. H.C. Khare, M.Sc., Ph.D. (McGill), F.N.A.Sc., Allahabad.
- 9. Prof. Pramod Tandon, Ph.D., F.N.A.Sc., Shillong.

Members

- 10. Prof. Asis Datta, Ph.D., D.Sc., F.N.A.Sc., F.A.Sc., F.N.A., New Delhi.
- 11. Prof. Girjesh Govil, Ph.D., F.N.A.Sc., F.A.Sc., F.N.A., F.T.W.A.S., Mumbai.
- 12. Dr. S.E. Hasnain, Ph.D., F.N.A.Sc., F.A.Sc., F.N.A. Hyderabad.
- 13. Dr. V.P. Kamboj, Ph.D., D.Sc., F.N.A.Sc., F.N.A., Lucknow.
- 14. Prof. C.L. Khetrapal, Ph.D., F.N.A.Sc., F.N A., Lucknow.
- 15 Dr. G.C. Mishra, Ph.D., F.N.A.Sc., Pune.
- 16. Dr S.P. Misra, M.D., D.M., F.A.C.G., F.R.C.P., F.N.A.Sc., Allahabad.
- 17. Prof. Dipendra Prasad, Ph.D., F.N.A.Sc., F.A.Sc., Allahabad.
- 18. Prof. K.S. Rai, Ph.D., F.N.A.Sc., Jalandhar.
- 19. Prof. Abhijit Sen, Ph.D., F.N.A.Sc., F.A.Sc., Gandhinagar.
- 20. Dr. (Mrs.) Manju Sharma, Ph.D., F.N.A.Sc., New Delhi.
- 21. Prof. U.S. Srivastava, M.Sc., M.Ed., D.Phil., D.I.C. (Lond.), F.N.A.Sc., F.N.A., Allahabad.
- 22. Prof. P.N. Tandon, M.S., F.R.C.S., F.A.M.S., F.N.A.Sc., F.A.Sc., F.N.A., F.T.W.A.S., Delhi.
- 23. Prof. M. Vijayan, Ph.D., F.N.A.Sc., F.A.Sc., F.N.A., Bangalore.

The *Proceedings of the National Academy of Sciences, India*, is published in two Sections: Section A (Physical Sciences) and Section B (Biological Sciences). Four parts of each section are published annually (since 1960).

The Editorial Board in its work of examining papers received for publication is assisted, in an honorary capacity by a large number of distinguished scientists. The Academy assumes no responsibility for the statements and opinions advanced by the authors. The papers must conform strictly to the rules for publication of papers in the *Proceedings*. A total of 25 reprints is supplied free of cost to the author or authors. The authors may ask for a reasonable number of additional reprints at cost price, provided they give prior intimation while returning the proof.

Communication regarding contributions for publication in the *Proceedings*, books for review, subscriptions etc. should be sent to the Managing Editor, The National Academy of Sciences, India, 5 Lajpatrai Road, Allahabad-211 002 (India).

Annual Subscription for both Sections: Rs. 500.00; for each Section Rs. 250.00; Single Copy: Rs. 100.00. Foreign Subscription: (a) for one Section: US \$100, (b) for both Sections U.S.\$ 200.

(Air-Mail charges included in foreign subscription)

Co-Sponsored by C.S.T., U.P. (Lucknow)

PROCEEDINGS

OF THE

NATIONAL ACADEMY OF SCIENCES, INDIA 2001

VOL. LXXI SECTION-A PART IV

Determination of sulphide sulphur in alkaline solutions

AKELLA SIVARAMAKRISHNA

Department of Engineering Chemistry, A.U. College of Engineering, Visakhapatnam-530 003, India.

Received July 24, 2000; Revised February 6, 2001; Accepted May 9, 2001

Abstract

Conditions have been developed for the indirect titrimetric determination of water soluble sulphide sulphur with ferricyanide as a reagent using copper (II) sulphate as catalyst. The excess ferricyanide content is titrated wih standard hydroxylamine solution using tris (4-7-dihydroxy-1, 10-phenanthroline) – Iron (III) as an indicator in strong alkali medium. The proposed method for the determination of sulphide sulphur is simple, rapid, and sensitive (0.3 to 20 mg). The pooled standard deviation and average error are found to be 0.04 mg and $\pm 0.44\%$ respectively.

(Keywords: titrimetry/sulphide sulphur/ferricyanide)

Introduction

The development of fairly effective analytical techniques becomes significant towards control methodologies in pollution studies. Sulphide sulphur is one such pollutant. As such a procedure that can be used to determine sulphide content with a reasonable degree of accuracy in the presence of some other sulphur compounds will have its own definite advantages. The various titrimetric methods reported for determining sulphide sulphur in different materials and waster waters were generally based on the reagents like iodine¹⁻⁴, permanganate⁵⁻⁶ and others⁷⁻¹⁰.

Most of these methods are usually associated with one or other drawbacks and limitations; for example, instability in acid solution, interference due to other sulphur compounds, requiring potentiometric assembly, high temperatures, which have been well

exemplified¹¹⁻¹². The reported methods¹²⁻¹⁴ for the determination of sulphide sulphur with ferricyanide under controlled pH in the absence of oxygen depend on its oxidation to elemental sulphur. Amperometric method reported needs the use of osmium tetroxide as catalyst and temperature while potentiometric methods have necessarily to be carried out at elevated temperatures. But so far no room temperature titrimetric visual end – point method could be found in the literature for the determination of sulphide sulphur based on its oxidation to the sulphate state with ferricyanide.

The present communication describes a rapid and convenient back titration method to the determination of sulphide using ferricyanide as oxidant. The procedure is based¹⁵ on the titration of excess ferricyanide content by hydroxylamine using tris (4, 7, dihydroxy-1, 10-phenanthroline)–Iron(III) [DHP)₃ – Iron(III)] as an indicator.

Materials and Method

All chemicals used were of analytical grade reagents. 0.1N solutions of potassium ferricyanide and hydroxylamine hydrochloride were prepared and standardized 16 . 0.25% copper (II) sulphate solution was prepared. 0.16% (DHP)₃ – Iron(III) indicator solution

Table 1 – Effect of Copper (II) Sulphate. Taken sulphide = 5.44 mg

Time of reaction (min.)	Sulphide found (mg)		% Recovery of sulphide		
	In the absence of Cu(II)	1.7 × 10 ⁻⁴ M Cu(II)	In the absence of Cu (II)	In the presence of 1.7×10^{-4} M Cu(II)	
1	3.64	5.20	66.9	96.0	
2	3.84	5.31	70.6	97.6	
3	4.08	5.43	75.0	99.8	
4	4.34	5.44	79.8	100.0	
6	4.84	5.39	88.9	99.1	
8	5.18	5.41	92.9	99.4	
10	5.38	5.38	98.9	98.9	
20	5.44	5.42	100.00	99.6	
30	5.42	5.44	96.6	100.0	

Table 2 - Determination of sulphide sulphur

Sulphide Sulphur (mg)						
Taken	Found					
	In pure samples	In the presence of 1.42mg sulphite	In the presence of 20.82 mg of thiosulphate			
0.450	0.454	0.456	0.454			
0.912	0.916	0.918	0.916			
2.736	2.726	2.728	2.730			
4.560	4.548	4.544	4.542			
9.120	9.149	9.142	91.53			

was prepared as reported¹⁷. Solutions of 0.01 M sodium sulphide (Na₂S.9H₂O), 0.05M sodium sulphite and 0.05M sodium thiosulphate were prepared and standardized¹². These solutions were prepared on the day of use. A 6N sodium hydroxide solution was prepared.

Take an aliquot of sulphide (0.3 to 20mg) and 30 ml of 6N sodium hydroxide into a conical flask. Add a known excess volume of ferricyanide and then 1ml of 0.25% copper (II) solution to it. Keep the reaction mixture aside for 3 min. Add 0.5ml of $(DHP)_3$ – Iron (III) indicator solution. At this stage the colour of solution is greenish yellow. Titrate the excess ferricyanide content with standard hydroxylamine solution until the colour changes from greenish yellow to intense red. A blank titration on the same ferricyanide volume is run.

Results and Discussion

Preliminary experiments in the determination of sulphide sulphur with ferricyanide revealed that the minimum time required for the complete oxidation of sulphide to sulphate is 10 min. The reaction time gets reduced to 3 min. in the presence of copper (II) as catalyst or temperature (50-60°C). The [Ferricyanide]/[Sulphide] should be greater than 89 to give results corresponding to the indicated accuracy. The order of addition of reagents should be strictly followed as given in the procedure. The results in Table 1 clearly show the positive catalyst effect of copper (II) on the rate of reaction between sulphide and ferricyanide. The stoichiometry of the reaction can be expressed by the following equation.

$$S^{2-} 8 [Fe(CN)_6]^{3-} + 8 OH^- \longrightarrow SO_4^{2-} + 8 [Fe(CN)_6]^{4-} + 4 H_2O$$

The addition of formaldehyde eliminates the interference of sulphite which forms an inert complex. 5-fold excess of sulphite and 100-fold excess of thiosulphate do not interfere with the determination.

The results of the proposed method for the determination of sulphide sulphur in synthetic mixtures are in good agreement with the standard method. The pooled standard deviation and average error of the present method are found to be 0.04mg and \pm 0.44% respectively. The results in the determination of sulphide are given in Table 2. Some of the advantages of the present method are :

- I. Does not require close control of pH.
- II. Thiosulphate and sulphite do not interfere
- III. Use of cheaper catalyst [Cu(II) as compared to Os (VIII)]

The method is more conveniently suitable for the estimation of sulphide sulphur in refinery spent caustic solutions as compared to the method described¹⁸.

Acknowledgements

This work belongs to a research project, funded by UGC, New Delhi (India). The author thanks Prof. Dr. U. Muralikrishna for his guidance and Miss. T.A.S.S. Santhisree for checking some of the results reported in this paper.

References

- 1. Murthy, A.R.V., Narayanan, V.A. & Rao, M.R.A. (1956) Analyst. 81: 373.
- 2. Kaushik, R.L. & Prasad, Rajendra (1969) J. Indian Chemical Society 46: 405.
- 3. Sugino, K. (1983) Bunseki Kagaku 32:752.
- 4. Hrnjic, R. (1984) Koza Obuca. 33:92.
- 5. Nasu, Y. & Nasu, T. (1985) Bunsekio, 7: 468.
- 6. Neglia, S. (1961) Clay Min. Bull. 4: 243.
- 7. Kabrt, L. & Burya, (1991) Petr. Technolopaliv. 59: 189.
- 8. Tan, C.J. Harrieder. M. Kissesel, G. Mahrwald, R. & Hennig, H. (1984) Acta hydrochim Hydrobiol. 12: 315.
- Aresentev, V.A., Borisor, B.N., Levlev, yu. v, Mashevskii. G.N. & Posypkin, L.P. (1983) Obogashch Rnd (Leningrade) 28: 21.
- 10. Lebska, B. & Krakowski, J. (1983) Przegl Paper. 39: 43.

- 11. Berka, A., Vulterin, J. & Zyka, J. (1965) Newer Redox Titrants. Pergamon Press Ltd., First Edition, Great Britain, p. 6, 24, 33, 49, 88.
- 12. Karchmer, J.H. (1970) *The Analytical chemistry of sulphur and its compounds*. Part I, Interscience, Vol. 29, p. 310.
- 13. Kiryushov, V.N. (1985) Zavod. Lab. 51:16.
- 14. Egorova, V.I., Vasileva, S.V. & Stolyarova, V.S. (1991) Zavod Lab. 57: 18.
- 15. Muralikrishna, U. & Sivaramakrishna, A. (1998) J. Chem. Env. Res. 7 (1&2): 157.
- 16. Vogel, A.I. (1989) A Text Book of Quantitative Chemical Analysis, ELBS, Longman, 5th Edition, Britain, p. 399, 408.
- 17 Poe, D.P. & Diehl, H. (1974) Talanta, 21: 1065.
- 18. Karchmer, J.H. & Dunahoe, J.W. (1948) Talanta 21:915.

Studies on oxovanadium (IV) and dioxouranium (VI) complexes with pyrazine-2-and pyrazine-2, 3-carboxylic acid hydrazide

G.S. SANYAL*, P.K. NATH and R. GANGULY

Department of Chemistry, University of Kalyani, Kalyani-741 235, India.

Received February 12, 2000; Final Revision March 16, 2001; Accepted May 31, 2001

Abstract

Complexes of oxovanadium (IV) and dioxouranium (VI) with pyrazine-2-carboxylic acid hydrazide (L) and pyrazine-2, 3-dicarboxylic acid dihydrazide (L') have been prepared and characterized through analytical, spectral and magnetic moment data. $VO(acac)_2$ with L gives oxovanadium (IV) complex of the Schiff's base generated in situ from the reaction of the hydrazide and 'acac' moiety; with L', VO(IV) and $UO_2(VI)$ result in isolation of complexes of pyrazine-3-hydrazide-2-carboxylic acid presumably derived in situ from the hydrolysis of a hydrazide group.

(Keywords: metal complexes/pyrazine carboxylic acid hydrazides/VO(IV)/UO₂(VI)/Schiff's base)

Introduction

Metal complexes of carboxylic acid hydrazides are interesting due to their biological and structural importance¹⁻³. Our previous investigation^{4.5} using pyrazine-2-carboxylic acid hydrazide (L) and pyrazine-2, 3-dicarboxylic acid dihydrazide (L') with several centres of coordination have prompted studies on their oxovanadium (IV) and dioxouranium (VI) halides, sulphates, acetates and acetylacetonates.

Vanadium is receiving considerable attention as a biologically important metal due to the existence of vanadoenzyme⁶. Dioxouranium (VI) complexes are of structural interest due to varied geometry they assume, of late dioxouranium (VI) complexes are said⁷ to have possible applications in solar energy conversion systems owing to spectral and excited state electron transfer properties of the UO_2^{2+} ion.

Materials and Method

Pyrazine-2-carboxylic acid hydrazide (L) and pyrazine-2, 3-dicarboxylic acid dihydrazide (L') (Fig. 1 & 2); $VO(acac)_2$ and $UO_2(acac)_2.2H_2O$; and $VOCl_2$ (ethanolic

296 G.S. SANYAL et al.

solution, pH ~ 3-4) were prepared following published procedures 4,5,8,9,10 . VOSO₄.H₂O and UO₂(CH₃COO)₂.2H₂O used were E Merck/BDH quality.

Spectral (IR, UV-Vis, NMR, EPR), magnetic moments and conductivity measurements and C, H, N analyses were done as described elsewhere ^{4,5}. Metal, halogen and sulphur were analysed gravimetrically by standard methods as oxides (V₂O₅, U₃O₈) and as AgCl and BaSO₄. Molecular weights were determined by Rast method with diphenyl as solvent. FAB-MS using Jeol SX-102 instrument were recorded in CDRI, Lucknow.

Complex $l: V_2O_5$ (0.470 g, ~2.58 mmol) was refluxed on a steam bath with dry ethanol (10 ml) containing 12 M HCl (5 ml). A clear green solution of VOCl₂ thus obtained was filtered in a basin and the filtrate concentrated to ~1 ml. The resultant solution (pH 3-4) after dilution with dry ethanol (5 ml) was mixed with solution of ligand L (145 g, ~10.51 mmol) in hot ethanol (50 ml). Immediate colour-change occurred and a red compound appeared. The suspension was refluxed (2h) on a steam bath. The red compound was filtered hot, washed with dry ethanol and dried over fused CaCl₂ in a desiccator. Yield: 60% m.p. 270°C (dec.).

Analy. data: found % (calc for VOLCl₂.4H₂O) M 14.04 (14.65), C 16.57 (17.24), H 4.17 (4.02), N 15.88 (16.09), Cl 18.60 (20.46). $\Lambda_{\rm M}$ (Ω^{-1} cm² mol⁻¹, DMSO): 146.1. $\mu_{\rm eff}$ (RT): 1.71 BM. EPR (RT): $g_{\rm av}$ = 1.97. IR (KBr), cm⁻¹: 3000-3500 (br), 1620(s), 1040 (ms), 970(s), 1240 (wsh), 1170-1140 (s), 1050-1030 (s); and UV-Vis (nujol-mull), cm⁻¹: 22730 (br), 19230 (br, sh), 13330 (uw).

Complex 2: A solution containing VOSO₄. H_2O (0.900 g, 5 mmol) in dry methanol (50 ml) was mixed with solution of ligand L (1.65 g, ~ 12 mmol) in the same solvent (50 ml) and the mixture was refluxed (4h) on a steam bath when a blackish compound formed was filtered hot and washed with hot methanol. The compound was purified by dissolving in hot water, filtering and evaporating the filtrate to dryness to obtain brown crystals which were dried in a desiccator over fused CaCl₂. Yield 56%, m.p. >300°C.

Analy. data: found % (calc. for VOLSO₄.2H₂O) M 14.81 (15.13), C19.13 (17.80), H 3.15 (2.97), N 16.75 (16.62), SO₄ 28.86 (28.48). $\Lambda_{\rm M}$ (Ω^{-1} cm² mol⁻¹, DMSO): 62.3. $\mu_{\rm eff}$ (RT): 1.69 BM. EPR (RT): $g_{\rm av}=1.96$ IR (KBr), cm⁻¹: 3000–3500 (br), 1620 (s), 1030 (m); and UV-Vis (nujol-mull) 23810 (br) cm⁻¹.

Complex 3: A mixed solution containing $VO(acac)_2$ (0.160 g, ~0.60 mmol) in dry methanol (10 ml) and solution of ligand L (0.165 g, ~1, 2 mmol) in the same solvent (10 ml) was refluxed (2h) on a steam bath. Brown coloured crystalline product separated was filtered hot, washed with hot methanol and dried over fused $CaCl_2$ in a desiscoator. Yield 60%, m.p. 240°C (dec.)

Analy. data: found % (calc. for (VO)₂ (acacL)₂) M 18.82 (17.89), C 41.34 (42.10), H 3.55 (3.51), N 19.63 (19.65). FAB-MS molecular peak observed at M/Z 570. $\Lambda_{\rm M}$ (Ω^{-1} , cm² mol⁻¹, DMSO): 46.6 $\mu_{\rm eff}$ (RT): 1.55 BM/V. EPR (RT): $g_{\rm av}$ = 1.96 IR (KBr), cm⁻¹: 1580 (s), 1420 (m), 530 (s), 840 (m, sh) 970 (s); and UV-Vis (nujol-mull), cm⁻¹: 22730 (br), 17860 (sh), 16670 (sh), 15150 (sh). *acacL = Schiff base (dianionic) from L and acac ion.

Complex 4: A mixed solution containing UO_2 (acac)2.2 H_2O (0.500 g, ~ .1 mmol) in dry methanol (20 ml) and ligand L (0.280 g, ~ 2.03 mmol) in the same solvent (20 ml) was refluxed (2h) on a steam bath, the precipitated brown complex was filtered hot washed with hot dry methanol and dried over fused $CaCl_2$ in a desiccator. Yield: 45%, m.p. >300°C.

Analy. data : found % (calc. for UO₂ (L-H)₂) M 44.50 (43.75), C 22.25 (22.05), H 2.09 (1.84), N 21.55 (20.59). $\Lambda_{\rm M}$ (Ω^{-1} cm² mol⁻¹, DMSO) : 26.0. IR (KBr), cm⁻¹; 1600(s), 1040 (ms), 12.80 (ms), 840 (s). 920 (s); UV-Vis (nujol-mull), cm⁻¹ 23260 (br), 16400 (sh); ¹H NMR, δ (ppm) : 5.08 (NH₂) (Fig. 3).

Complex 5: To a solution (1 ml) of $VOCl_2$ prepared from V_2O_5 (0.960 g, 5.18 mmol) as a complex 1 and diluted with dry methanol (10 ml) was added to suspension of ligand L' (1.01 g, ~5.15 mmol) in dry methanol (100 ml) and the whole refluxed (3h). The suspension became clear and was filtered. The filtrate was concentrated (1/2 v), refrigerated and then mixed with a little ether. Brown complex formed was filtered, washed with cold methanol and dried over fused $CaCl_2$ in a desiccator. Yield 40%, m.p. 290°C (dec.).

Analy. data : found % (calc. for (VO)2L' (OCH₃)₂Cl₂.2H₂O) M 19.13 (20.44), C21.55 (19.24), H 3.84 (3.61), N 16.97 (16.83), Cl 15.90 (14.23). $\Lambda_{\rm M}$ (Ω^{-1} cm² mol⁻¹, DMSO) : 97.1. $\mu_{\rm eff}$ (RT) : 1.70 BM/V. EPR (RT) : $g_{\rm av}$ = 197. IR (KBr), cm⁻¹ : 2500–3500 (br), 1650

298 G.S. SANYAL et al

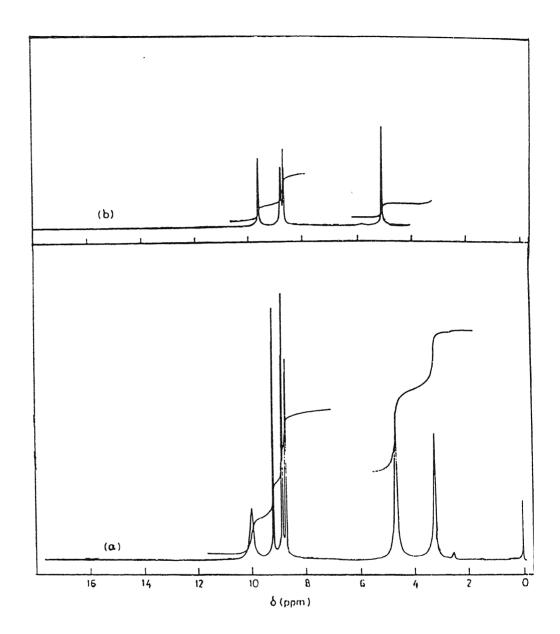


Fig. $3 - {}^{1}HNMR$ spectra of (a) L and (b) complex 4 in DMSO-d₆ (Ref. TMS)

(intensity largely reduced), 450-460 (ms); and UV-Vis (nujol-mull), cm⁻¹: 25000 (br), 13890 (vw).

Complex 6: A solution of VO (acac)₂ (1.2 g, \sim 4.53 mmol) in dry methanol (50 ml) was mixed with a suspension of ligand L' (0.900 g \sim 4.60 mmol) in dry methanol (100 ml) and the whole refluxed (5h) on a steam bath when the suspension became clear. The solution was filtered and the filtrate was concentrated (1/2 v), refrigerated and then mixed with a little ether. Ash coloured complex that settled down was filtered, washed with chilled methanol and dried over fused CaCl₂ in a desiccator. Yield: 38% m.p. > 300°C.

Analy. data : (calc. for (VO)₂ (OH)₂ (Pz-3-CONHNH₂-2-COO)₂) M 20.00 (19.24), C27.67 (27.17), H 2.71 (2.26), N 20.24 (21.13). $\Lambda_{\rm M}$ (Ω^{-1} cm² mol⁻¹, DMSO) : 31.9 $\mu_{\rm eff}$ (RT) : 1.17 BM/V EPR (RT) : $g_{\rm av}$ = 1.97 IR (KBr), cm⁻¹ : 1650 (s), 1590 (ms), 1330 (ms), 965 (s), 460 (s), 580 (m); and UV-Vis (nujol-mull), 25000 (br) cm⁻¹.

Complex 7: A solution of $VOSO_4$. H_2O (0.180 g, ~1.10 mmol) in dry methanol (20 ml) was mixed wth a solution of ligand L' (0.450 g, 2.29 mmol) in hot water-methanol (1.3 v/v, 30 ml) and the mixture refluxed (5h) on a steam bath. The resulting yellow-green precipitate was filtered hot, washed with hot aq. methanol and dried over fused $CaCl_2$ in a dessicator Yield: 40% m.p. > 300°C.

Analy. data: found % (calc. for (VO)₂(OH)₂L'₂(SO₄)₄H₂O) M 14.04 (14.01), C 21.65 (19.78), H 3.22 (3.57), N 22.43 (23.08) $\Lambda_{\rm M}$ (Ω^{-1} cm² mol⁻¹, DMSO): 47.0 $\mu_{\rm eff}$ (RT): 1.30 BM/V. EPR (RT): $g_{\rm av}$ = 1.96 IR (KBr), cm⁻¹: 3000-3500 (br), 1650 (intensity largely reduced), 1100 (s) 610 (s), 450-460 (ms), 970 (s); and UV-Vis (nujol-mull): 23530 (w) cm⁻¹.

Complex 8: A solution of UO_2 (CH₃COO)₂ 2H₂O (0.210 g, ~0.5 mmol) in hot water-methanol (1:3 v/v, 20 ml) was mixed with solution of ligand L' (0.200 g, ~1.02 mmol) in same hot solvent mixture (30 ml) and the contents refluxed (2h) on a steam bath. The resulting red compound formed was filtered hot, washed with hot methanol and dried over fused CaCl₂ in a desisccator. Yield 60%, m.p. > 300°C

Analy. data: found % (calc. for $UO_2(Pz-3-CONHNH_2-2COO)_2$) M 37.67 (37.66), C 22.45 (22.78), H 1.68 (1.58), N 16.41 (17.72). Λ_M (Ω^{-1} cm² mol⁻¹, DMSO): 30.8, IR (KBr), cm⁻¹: 1330 (s), 1590 (s), 1130 (ms), 450-460 (ms); and UV-Vis (nujol-mull): 22730 (br) cm⁻¹, 1H NMR, δ(ppm): 942-1024 (Pz-ring), 11.97 (NH).

Results and Discussion

Conductivity values of 10⁻³ M solution in DMSO for complexes other than 1, 2 and 5 are less for 1:1 electrolytes and suggest their nonelectrolytic behaviour despite some solvolysis. Molecular weight measurements indicate complexes 3, 6 and 7 are dimers.

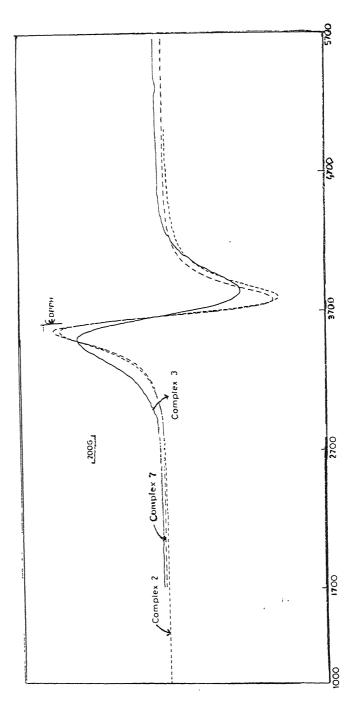


Fig. 4- EPR spectra (RT) frequency 9.45GHz. Field set 3000G for complex 2 and 3700G for complexes 3&7.

Infrared spectra: Pyrazine-2-carboxylic acid hydrazide (L) shows bands at 1675 and 1020 cm^{-1} tentatively assigned to VC = O and VN-N respectively; these suffered negative and positive shifts ($\Delta v \sim 20 \text{ cm}^{-1}$) in complexes 1 and 2 suggesting coordination through the carboxyl oxygen and the terminal hydrazinic nitrogen. Bands observed for 2 in the region 1240 (w, sh), 1170-1139, 1050-1030 cm⁻¹ indicate 12 sulphate is bidentately chelated. In complex 3, disappearance for vC=O and appearance of a very strong band to around 1580 cm⁻¹ are characteristic of >C=N-N=C< group. A new band at ~1420 cm⁻¹ (absent in the free ligand) may be assigned 12 to deformation mode of methyl group derived from 'acac' moiety. Non-ligand bands at 530(s) and 840 cm⁻¹ (m, sh) may be attributed 13 to O-V-O sym and asym stretching modes of the oxobridged portion of the dimer. In complex 4, vC=O band disappeared while a strong band appeared at ~1600 cm⁻¹. This along with a new band at 1260 cm⁻¹ suggests¹ coordination through the deprotonated enol oxygen. Positive shift ($\Delta v \sim 20 \text{ cm}^{-1}$) of vN-N band in 4 points to coordination through the terminal hydrazinic nitrogen. Absence of N-H signal from its free ligand position at δ 10 ppm suggests deprotonation via enolisation, and downfield shift of free ligand signal at δ 4.64 ppm to δ 5.08 ppm in the complex supports coordination through NH₂ group of the ligand (Fig. 3).

In complexes 5 and 7, the strong vC=O band at 1650 cm⁻¹ of the free dihydrazide ligand (L') is reduced in intensity (though not shifted); coordination of keto oxygen to the metal is indicated. Band attributable 12 to C-O stretching of coordinated methoxide in 5 at around 1000 cm⁻¹ is not, however, discernible from the complexity of spectra. Sulphate is ionic in 7 apparent from the strong band around 1110 cm⁻¹ and another at 610 (ms) cm⁻¹. In complexes 5-8, coordination through heterocylic ring nitrogen is suggested from the observed shift of pyrazine ring out-of-plane deformation band to higher frequencies (450-460 cm⁻¹) relative to the free ligand (400, 415 cm⁻¹). Downfield shift of protons of pyrazine ring from the region $\delta 8.63 - 9.08$ to $\delta 9.42 - 10.24$ ppm in the ¹H NMR spectrum of 8 indicates coordination of pyrazine ring nitrogen to the metal: the signal at δ 11.97 ppm is ascribed to NH (perhaps hydrogen bonded) of a free hydrazide residue. Complexes 6 and 7 which are dimers are possibly hydroxo-bridged, a new band (ms) around 560-580 cm⁻¹ is assignable 14 to v(V-O) of the μ -hydroxobridges. A strong characteristic band for complexes 6 and 8 at ~1330 cm⁻¹ (absent in the free ligand) is perhaps due to $v_{as}COO$; the corresponding v_{as} COO may be assigned to the strong band at ~1590 cm⁻¹. The complexes show a free amide I band (strong) at ~1650 cm⁻¹; presence of a free hydrazide group is evident¹¹ from NH₂ rocking mode ~1130 cm⁻¹.

In all the complexes strong band assignable to vV=O around 970-980 cm $^{-1}$; and to v_{as} O=U=O in the region 880-910 cm $^{-1}$ have been observed. Non-ligand bands occurring in the regions 500, 550 and 370 cm $^{-1}$ may be assigned 12 , respectively, to U-O and U-N

302 G.S. SANYAL et al.

vibrations. Broad band observed in the region 3000-3500 cm⁻¹ for complexes 1, 2, 5 and 7 suggest presence of water and/or hydrogen bonding. Isothermal (110°C, air oven) mass loss (9.57, 7.19 and 10.6%) for the corresponding complexes 2, 5 and 7 suggests crystalline water entirely and that (11.5%) for 1 partly (2 mols).

Magnetic moment (μ_{eff}) values observed at room temperature (33°C) for the oxovanadium (IV) complexes (1, 2 and 5) are normal for magnetically dilute species. Complexes 3, 6 and 7 give lower values similar to those observed lise elsewhere, lowering of magnetic moment in these dimers may indicate antiferromagnetic coupling between the vanadium centers since fortunately, spin-orbit mooupling in nearly all V(IV) complexes is efficiently quenched le.

EPR spectra of the oxovanadium (IV) complexes in polycrystalline state at room temperature (Fig. 4) are nearly isotropic and the g_{av} values are close to 1.96-1.97. No half-filled transition has been located in polycrystalline solids at room temperature even at high gain (however, in absence of epr study at low temperature and/or variable temperature magnetic measurements, which are inaccessible to us, results remain inconclusive).

Electronic spectra of the oxovanadium (IV) complexes in nujol mull exhibit bands in the region 400-440 and 500-600 nm attributable¹⁵, respectively to ${}^2B_2(d_{xy}) \rightarrow {}^2A_1(d_{z2})$, ${}^2B_2(d_{xy}) \rightarrow {}^2B_1(d_{x2-y2})$ transitions.

Non-ligand bands observed for complexes 4 and 8 in the region 430-400 nm may be assigned to the $^{1}\sum_{\nu}^{+} \rightarrow ^{3}\Pi_{\nu}$ transition of the UO₂ moiety 17 .

Acknowledgements

The authors thank the authorities of CDRI, Lucknow, for FAB-MS and RSIC, Bose Institute, Kolkata, for EPR measurements. One of the authors (RG) thanks University of Kalyani for the award of Fellowship.

References

- 1. Dutta, R.L. & Hossain, M.M. (1985) J. Sci. Ind. Res. 44: 635.
- Albert, A. (1956) Nature 177: 525.
- 3. Kogan, V.A., Zelenstov, V.V., Gerbelev, N.V. & Lukov, V.V. (1986) Russ. J. Norg. Chem. 31: 1628.
- 4. Sanyal, G.S. & Garai, S. (1991) Indian J. Chem. 30A: 554 and references therein.
- 5. Sanyal, G.S., Garai, S. & Banerjee, P. (1990) Synth. React. Inorg. Met.-Chem. 23: 559.
- Frausto, da Silva J.J.R. & Williams, R.J.P. (1991) The biological chemistry of the elements. The inorganic chemistry of life. Clarendrow Press, Oxford, p. 427.

- 7. Rosenfeld-Grunwald, T. & Bonini, J. (1980) J. Phy. Chem. 84: 2981 and references therein.
- 8. Fernelius, W.C. & Bryant, B.E. (1957) Inorg. Synth. 5:105.
- 9. Chowdhury, M.K., Paul, P.C. & Srinivas, P. (1992) Proc. Indian Acad. Sci. (Chem. Sci.) 104: 479.
- 10. McCormick, B.J. Featherstone, J.L., Stoklosa, H.J. & Wasson, J.R. (1973) Inorg. Chem. 12:692.
- 11. Aggarwal, R.C. & Rao, T.R. (1977) Current Sci. 46: 625.
- 12. Nakamoto, K. (1970) Infrared spectra of inorganic & coordination compounds, 2nd Edition, John Wiley and Sons, New York.
- 13. Cavaco, I., Pessoa, J.C., Duarte, M.T., Henriques, R.T., Matias, P.M. & Gillard, R.D. (1996) J. Chem. Soc. Dalton Trans. 1989.
- 14. Wieghardt, K., Bossek, U., Volckmar, K., Swiridoff, W. & Weiss, J. (1984) Inorg. Chem. 23: 1387.
- 15. Syamal, A. (1975) Coord. Chem. Rev. 16:309 and references therein.
- 16. Carrano, C.J., Nunn, C.M., Quan, R., Bonadies, J.A. & Pecoraro, V.L. (1990) Inorg. Chem. 29: 944.
- 17. McGlynn, S.P. & Smith, J.K. (1961) J. Mol. Spectros. 6: 169.

Reflection and transmission of P and SV waves at the interface between two monoclinic elastic half-spaces

SARVA JIT SINGH and SANDHYA KHURANA

Department of Mathematics, Maharshi Dayanand University, Rohtak-124 001, India.

Received December 6, 1999; Accepted April 2, 2000

Abstract

Reflection and transmission of quasi -P (qP) and quasi -SV (qSV) waves at the interface between two monoclinic elastic half-spaces is discussed. Closed-form expressions for the reflection and transmission coefficients are derived. A method of computing these coefficients is indicated. The present analysis corrects some fundamental errors appearing in a recent paper on the reflection and transmission of qP waves at the interface between two monoclinic media.

(Keywords: reflection/transmission/waves/monoclinic media)

Introduction

Musgrave¹ discussed the reflection and transmission of elastic waves at a plane boundary between two anisotropic media of hexagonal type. Dayley and Hron^2 investigated the case when the media involved are transversely isotropic. Keith and Crampin³ derived a formulation for computing the energy division among qP, qSV and qSH waves generated by plane waves incident on a plane boundary between generally anisotropic media. A comprehensive account was presented for the case of isotropic/orthotropic interface.

A monoclinic medium possesses one plane of elastic symmetry. For wave propagation in the plane of symmetry, SH motion is decoupled from the P-SV motion. While the particle motion of SH waves is purely transverse, it is neither purely longitudinal nor purely transverse in the case of P-SV waves. In a recent paper, Chattopadhyay and Sha^4 discussed the reflection of qP waves at the interface between two monoclinic half-spaces. Since the authors assume that qP waves are purely longitudinal and qSV waves purely transverse, most of the results of this paper, including the expressions for the reflection and transmission coefficients, are erroneous (see also $Singh^5$). The aim of the present study is to derive closed-form algebraic expressions for the reflection and transmission coefficients when plane waves of qP or qSV type are incident at the plane boundary between two monoclinic elastic half-spaces. A method of computing the reflection and transmission coefficients is indicated. Numerical results will be presented in a subsequent publication.

Plane Waves in a Monoclinic Elastic Medium

Consider a homogeneous anisotropic elastic medium of monoclinic type. It has one plane of elastic symmetry and its elastic properties are defined by thirteen elastic moduli. Taking the plane of symmetry as the x_2x_3 -plane, the generalized Hooke's law can be expressed in the form

$$\tau_{11} = c_{11} e_{11} + c_{12} e_{22} + c_{13} e_{33} + 2c_{14} e_{23}, \tag{1a}$$

$$\tau_{22} = c_{12} e_{11} + c_{22} e_{22} + c_{23} e_{33} + 2e_{24} e_{23}, \tag{1b}$$

$$\tau_{33} = c_{13} e_{11} + c_{23} e_{22} + c_{33} e_{33} + 2c_{34} e_{23}, \tag{1c}$$

$$\tau_{23} = c_{14} e_{11} + c_{24} e_{22} + c_{34} e_{33} + 2c_{44} e_{23}, \tag{1d}$$

$$\tau_{13} = 2 \left(c_{55} \, e_{13} + c_{56} \, e_{12} \right), \tag{1e}$$

$$\tau_{12} = 2 \left(c_{56} e_{13} + c_{66} e_{12} \right), \tag{1f}$$

where τ_{ii} is the stress tensor and e_{ii} the strain tensor. Further,

$$2e_{ij} = \frac{\partial u_i}{\partial x_i} + \frac{\partial u_j}{\partial x_i},\tag{2}$$

u, being the displacement vector.

For plane waves propagating in the x_2x_3 -plane

$$u_i = u_i(x_2, x_3, t), \ \partial/\partial x_1 \equiv 0. \tag{3}$$

The equations of motion without body forces are

$$\frac{\partial}{\partial x_i} \tau_{ij} = \rho \frac{\partial^2 u_i}{\partial t^2} (i = 1, 2, 3), \qquad (4)$$

using the summation convention. From eqn. (1) to (4), we obtain the equations of motion in terms of the displacements in the form

$$c_{66} \frac{\partial^2 u_1}{\partial x_2^2} + 2c_{56} \frac{\partial^2 u_1}{\partial x_2 \partial x_3} + c_{55} \frac{\partial^2 u_1}{\partial x_3^2} = \rho \frac{\partial^2 u_1}{\partial t^2}, \tag{5}$$

$$c_{22} \; \frac{\partial^2 u_2}{\partial x_2^2} \; + \; c_{44} \; \frac{\partial^2 u_2}{\partial x_3^2} \; + \; c_{24} \; \frac{\partial^2 u_3}{\partial x_2^2} \; + \; c_{34} \; \frac{\partial^2 u_3}{\partial x_3^2} \; + \; 2c_{24} \; \frac{\partial^2 u_2}{\partial x_2 \, \partial x_3}$$

$$+ (c_{23} + c_{44}) \frac{\partial^2 u_3}{\partial x_2 \partial x_3} = \rho \frac{\partial^2 u_2}{\partial t^2},$$
 (6)

$$c_{24} \frac{\partial^{2} u_{2}}{\partial x_{2}^{2}} + c_{34} \frac{\partial^{2} u_{2}}{\partial x_{3}^{2}} + c_{44} \frac{\partial^{2} u_{3}}{\partial x_{2}^{2}} + c_{33} \frac{\partial^{2} u_{3}}{\partial x_{3}^{2}} + 2c_{34} \frac{\partial^{2} u_{3}}{\partial x_{2} \partial x_{3}}$$

$$+ (c_{23} + c_{44}) \frac{\partial^2 u_2}{\partial x_2 \partial x_3} = \rho \frac{\partial^2 u_3}{\partial t^2}.$$
 (7)

From eqn. (5) to (7), it is obvious that the u_1 motion representing SH waves is decoupled from the (u_2, u_3) motion representing qP and qSV waves.

Let $p(0, p_2, p_3)$ denote the unit propagation vector, c the phase velocity and k the wave number of plane waves propagating in the x_2x_3 -plane. We seek plane wave solutions of the equations of motion (6) and (7) of the form

$$\begin{pmatrix} u_2 \\ u_3 \end{pmatrix} = A \begin{pmatrix} d_2 \\ d_3 \end{pmatrix} \exp \left[ik \left(ct - x_2 p_2 - x_3 p_3\right)\right], \tag{8}$$

where $d(0, d_2, d_3)$ is the unit displacement vector, also known as the polarization vector. Inserting the expressions for u_2 and u_3 in the equations of motion (6) and (7), we obtain

$$(U - \rho c^2) d_2 + V d_3 = 0, (9)$$

$$V d_2 + (Z - \rho c^2) d_3 = 0, (10)$$

where

$$U(p_2, p_3) = c_{22}p_2^2 + c_{44}p_3^2 + 2c_{24}p_2p_3,$$

$$V(p_2, p_3) = c_{24}p_2^2 + c_{34}p_3^2 + (c_{23} + c_{44})p_2p_3,$$

$$Z(p_2, p_3) = c_{44}p_2^2 + c_{33}p_3^2 + 2c_{34}p_2p_3.$$
(11)

Eqn. (9) and (10) yield

$$d_2/d_3 = V/(\rho c^2 - U) = (\rho c^2 - Z)/V.$$
 (12)

Therefore, ρc^2 satisfies the quadratic equation

$$\rho^2 c^4 - (U + Z) \rho c^2 + (UZ - V^2) = 0, \tag{13}$$

with solutions

$$2\rho c^{2}(p_{2}, p_{3}) = (U+Z) \pm [(U-Z)^{2} + 4V^{2}]^{1/2}.$$
 (14)

The upper sign in eqn. (14) is for qP waves and the lower sign is for qSV waves.

It has been shown by Singh⁵ that eqn. (8) will represent a pure longitudinal or transverse wave if

$$c_{24} p_2^4 + (c_{23} - c_{22} + 2c_{44}) p_2^3 p_3 - 3(c_{24} - c_{34}) p_2^2 p_3^2$$
$$- (c_{23} - c_{33} + 2c_{44}) p_2 p_3^3 - c_{34} p_3^4 = 0.$$
(15)

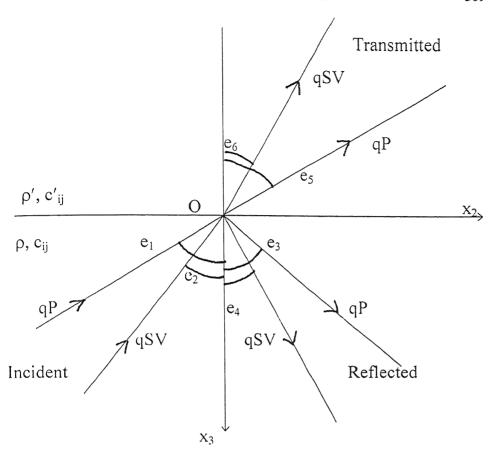


Fig. 1 – Reflection and transmission of qP and qSV waves at the plane interface ($x_3 = 0$) between two monoclinic half-spaces.

Eqn. (15) gives the directions of propagation for which P waves are purely longitudinal and SV waves purely transverse.

Reflection and Transmission of qP and qSV Waves

Consider a homogeneous, monoclinic, elastic half-space occupying the region $x_3 > 0$ in welded contact with another homogeneous, monoclinic, elastic half-space $x_3 < 0$ (Fig. 1). The identical plane of elastic symmetry of the two media is taken as the x_2x_3 - plane. Plane qP or qSV waves are incident at the interface $x_3 = 0$ from the half-space $x_3 > 0$. We consider plane strain problem for which

$$u_1 = 0, \ u_2 = u_2(x_2, x_3, t), \ u_3 = u_3(x_2, x_3, t).$$
 (16)

Incident qP or qSV waves will generate reflected qP and qSV waves in the half-space $x_3 > 0$ and transmitted qP and qSV waves in the half-space $x_3 < 0$. The total displacement field is given by

$$u_2 = \sum_{j=1}^4 A_j e^{iP_j}, \qquad u_3 = \sum_{j=1}^4 B_j e^{iP_j}$$
 (17)

for $x_3 > 0$, and

$$u_2' = \sum_{i=5}^{6} A_j e^{iP_j}, \qquad u_3' = \sum_{i=5}^{6} B_j e^{iP_j}$$
 (18)

for $x_3 < 0$, where

$$P_i = \omega \left[t - (x_2 \sin e_i - x_3 \cos e_i) / c_i \right], \quad (j = 1, 2, 5, 6; \text{ no summation})$$
 (19)

$$P_{j} = \omega \left[t - (x_{2} \sin e_{j} + x_{3} \cos e_{j}) / c_{j} \right], \qquad (j = 3, 4)$$
(20)

 ω being the angular frequency. We distinguish quantities corresponding to various waves by using the subscript (1) for incident qP waves, (2) for incident qSV waves, (3) for reflected qP waves, (4) for reflected qSV waves, (5) for transmitted qP waves and (6) for transmitted qSV waves. Thus, for example, for the incident qP waves, c_1 denotes the phase velocity, e_1 the angle of incidence, $P_1(x_2, x_3, t)$ the phase factor, A_1 the amplitude factor of the u_2 component of the displacement and B_1 that of the u_3 component. The displacement components in the half-space $x_3 < 0$ are denoted by u_2' and u_3' .

Since each of the incident qP, incident qSV, reflected qP, reflected qSV, transmitted qP and transmitted qSV waves must satisfy the equations of motion, we have, as in eqn. (12) and (14),

$$A_i = F_i B_i \ (i = 1, 2, ..., 6),$$
 (21)

where

$$F_i = V_i / (\rho c_i^2 - U_i) = (\rho c_i^2 - Z_i) / V_i, \quad (i = 1, 2, 3, 4),$$
 (22a)

$$F_i = V_i / (\rho' c_i^2 - U_i) = (\rho' c_i^2 - Z_i) / V_i$$
, $(i = 5, 6)$, (22b)

$$2\rho c_i^2 = (U_i + Z_i) + [(U_i - Z_i)^2 + 4V_i^2]^{1/2}, \quad (i = 1, 3),$$
(23a)

$$2\rho c_i^2 = (U_i + Z_i) - [(U_i - Z_i)^2 + 4V_i^2]^{1/2}, \quad (i = 2, 4)$$
(23b)

$$2\rho' c_5^2 = U_5 + Z_5 + [(U_5 - Z_5)^2 + 4V_5^2]^{1/2},$$
(24a)

$$2\rho' c_6^2 = U_6 + Z_6 - [(U_6 - Z_6)^2 + 4V_6^2]^{1/2}.$$
 (24b)

The expressions for U_i , V_i and Z_i are obtained from the expressions for U, V and Z given in eqn. (11) on substituting suitable values for (p_2, p_3) . For incident qP waves, $p_2 = \sin e_1$, $p_3 = -\cos e_1$; for incident qSV waves, $p_2 = \sin e_2$, $p_3 = -\cos e_2$; for reflected qP waves, $p_2 = \sin e_3$, $p_3 = \cos e_3$; for reflected qSV waves, $p_2 = \sin e_4$, $p_3 = \cos e_4$; for transmitted qP waves, $p_2 = \sin e_5$, $p_3 = -\cos e_5$; and, for transmitted qSV waves, $p_2 = \sin e_6$, $p_3 = -\cos e_6$ (see Fig. 1). We thus obtain

$$U_1 = c_{22} \sin^2 e_1 + c_{44} \cos^2 e_1 - 2c_{24} \sin e_1 \cos e_1,$$

$$V_1 = c_{24} \sin^2 e_1 + c_{34} \cos^2 e_1 - (c_{23} + c_{44}) \sin e_1 \cos e_1$$
,

$$Z_1 = c_{44} \sin^2 e_1 + c_{33} \cos^2 e_1 - 2c_{34} \sin e_1 \cos e_1;$$
 (25)

$$U_3 = c_{22} \sin^2 e_3 + c_{44} \cos^2 e_3 + 2c_{24} \sin e_3 \cos e_3$$
,

$$V_3 = c_{24} \sin^2 e_3 + c_{34} \cos^2 e_3 + (c_{23} + c_{44}) \sin e_3 \cos e_3$$
,

$$Z_3 = c_{44} \sin^2 e_3 + c_{33} \cos^2 e_3 + 2c_{34} \sin e_3 \cos e_3; \tag{26}$$

$$U_5 = c'_{22} \sin^2 e_5 + c'_{44} \cos^2 e_5 - 2c'_{24} \sin e_5 \cos e_5$$

$$V_5 = c'_{24} \sin^2 e_5 + c'_{34} \cos^2 e_5 - (c'_{23} + c'_{44}) \sin e_5 \cos e_5$$

$$Z_5 = c'_{44} \sin^2 e_5 + c'_{33} \cos^2 e_5 - 2c'_{34} \sin e_5 \cos e_5.$$
 (27)

 (U_2, V_2, Z_2) are obtained from (U_1, V_1, Z_1) on replacing e_1 by e_2 , (U_4, V_4, Z_4) are obtained from (U_3, V_3, Z_3) on replacing e_3 by e_4 and (U_6, V_6, Z_6) are obtained from (U_5, V_5, Z_5) on replacing e_5 by e_6 .

The total displacement field given by eqn. (17) and (18) must satisfy the boundary conditions.

$$u_2 = u'_2, u_3 = u'_3, \tau_{23} = \tau'_{23}, \tau_{33} = \tau'_{33} \text{ at } x_3 = 0.$$
 (28)

Since the boundary conditions (28) are to be satisfied for all values of x_2 , we must have

$$P_1(x_2, 0) = P_2(x_2, 0) = P_3(x_2, 0) = P_4(x_2, 0) = P_5(x_2, 0) = P_6(x_2, 0).$$
 (29)

Eqn. (19), (20) and (29) imply

$$\frac{\sin e_1}{c_1(e_1)} = \frac{\sin e_2}{c_2(e_2)} = \frac{\sin e_3}{c_3(e_3)} = \frac{\sin e_4}{c_4(e_4)} = \frac{\sin e_5}{c_5(e_5)} = \frac{\sin e_6}{c_6(e_5)} = \frac{1}{c_a},$$
 (30)

where c_a is the apparent phase velocity. This is the form of Snell's law for monoclinic media.

From eqn. (3a), (25) and (26), we note that even if $e_1 = e_3$, $c_1 \neq c_3$. Therefore, from eqn. (30), the angle of reflection of qP waves is not equal to the angle of incidence of qP waves. Similarly, the angle of reflection of qSV waves is not equal to the angle of

idence of qSV waves. Chattopadhyay and Saha⁴ assume that the angle of reflection of (qSV) waves is equal to the angle of incidence of qP (qSV) waves. Therefore, the lection and the transmission coefficients obtained by Chattopadhyay and Saha⁴ are not rect.

Using the relations (21), (29) and (30) the bounday conditions (28) yield

$$B_1 + B_2 + B_3 + B_4 - B_5 - B_6 = 0,$$
 (31a)

$$F_1B_1 + F_2B_2 + F_3B_3 + F_4B_4 - F_5B_5 - F_6B_6 = 0,$$
 (31b)

$$a_1B_1 + a_2B_2 + a_3B_3 + a_4B_4 - a_5B_5 - a_6B_6 = 0,$$
 (31c)

$$b_1B_1 + b_2B_2 + b_3B_3 + b_4B_4 - b_5B_5 - b_6B_6 = 0$$
, (31d)

iere

$$a_{1} = c_{24}F_{1} + c_{44} - (c_{44}F_{1} + c_{34})\cot e_{1},$$

$$a_{2} = c_{24}F_{2} + c_{44} - (c_{44}F_{2} + c_{34})\cot e_{2},$$

$$a_{3} = c_{24}F_{3} + c_{44} + (c_{44}F_{3} + c_{34})\cot e_{3},$$

$$a_{4} = c_{24}F_{4} + c_{44} + (c_{44}F_{4} + c_{34})\cot e_{4},$$

$$a_{5} = c'_{24}F_{5} + c'_{44} - (c'_{44}F_{5} + c'_{34})\cot e_{5},$$

$$a_{6} = c'_{24}F_{6} + c'_{44} - (c'_{44}F_{6} + c'_{34})\cot e_{6},$$

$$b_{1} = c_{23}F_{1} + c_{34} - (c_{34}F_{1} + c_{33})\cot e_{1},$$

$$b_{2} = c_{23}F_{2} + c_{34} - (c_{34}F_{2} + c_{33})\cot e_{2},$$

$$b_{3} = c_{23}F_{3} + c_{34} + (c_{34}F_{3} + c_{33})\cot e_{3},$$

$$b_4 = c_{23}F_4 + c_{34} + (c_{34}F_4 + c_{33})\cot e_4,$$

$$b_5 = c'_{23}F_5 + c'_{34} - (c'_{34}F_5 + c'_{33})\cot e_5,$$

$$b_6 = c'_{23}F_6 + c'_{34} - (c'_{34}F_6 + c'_{33})\cot e_6.$$

Incident qP waves:

In the case of incident qP waves, $A_2 = B_2 = 0$ and A_1 , B_1 are supposed to be known. Eqn. (31a, b, c, d) then constitute a set of four simultaneous equations in four unknowns, namely, B_3 , B_4 , B_5 and B_6 . These equations can be solved by Cramer's rule. We find

$$B_i/B_1 = \Delta_i^p/\Delta \quad (i = 3, 4, 5, 6),$$
 (32)

where Δ and Δ_i^p are defined in Appendix A. Using eqn. (21), we find

$$\frac{A_i}{A_1} = \frac{F_i}{F_1} \left(\frac{B_i}{B_1} \right) = \frac{F_i}{F_1} \left(\frac{\Delta_i^p}{\Delta} \right) (i = 3, 4, 5, 6; \text{ no summation over } i). \tag{33}$$

Incident qSV waves:

For incident qSV waves, $A_1 = B_1 = 0$ and A_2 , B_2 are supposed to be known. The amplitude ratios are found to be

$$B_i/B_2 = \Delta_i^S/\Delta, \tag{34}$$

$$\frac{A_i}{A_2} = \frac{F_i}{F_2} \left(\frac{B_i}{B_2} \right) = \frac{F_i}{F_2} \left(\frac{\Delta_i^S}{\Delta} \right) \qquad (i = 3, 4, 5, 6), \tag{35}$$

where Δ_i^S are defined in Appendix A.

Isotropic half-spaces:

For an isotropic medium,

$$c_{11} = c_{22} = c_{33} = \lambda + 2\mu$$

$$c_{12} = c_{13} = c_{23} = \lambda$$
, $c_{44} = c_{55} = c_{66} = \mu$,
 $c_{14} = c_{24} = c_{34} = c_{56} = 0$, (36)

where λ , μ are the Lamé parameters. Using these values for c_{ij} and similar values for c'_{ij} we obtain

$$c_{1} = c_{3} = \left[(\lambda + 2\mu) / \rho \right]^{1/2} = \alpha, \quad c_{2} = c_{4} = (\mu / \rho)^{1/2} = \beta,$$

$$c_{5} = \left[(\lambda' + 2\mu') / \rho' \right]^{1/2} = \alpha', \quad c_{6} = (\mu' / \rho')^{1/2} = \beta',$$

$$e_{1} = e_{3} = e, \quad e_{2} = e_{4} = f, \quad e_{5} = e', \quad e_{6} = f',$$

$$\frac{\sin e}{\alpha} = \frac{\sin f}{\beta} = \frac{\sin e'}{\alpha'} = \frac{\sin f'}{\beta'},$$

$$F_{1} = -F_{3} = -\tan e, \quad F_{2} = -F_{4} = \cot f, \quad F_{5} = -\tan e', \quad F_{6} = \cot f',$$

$$a_{1} = a_{3} = 2\mu, \quad a_{2} = a_{4} = -\mu \cos 2f / \sin^{2}f, \quad a_{5} = 2\mu', \quad a_{6} = -\mu' \cos 2f' / \sin^{2}f',$$

$$b_{1} = -b_{3} = -2\mu (\alpha/\beta)^{2} \cos 2f / \sin 2e,$$

$$b_{2} = -b_{4} = -2\mu \cot f, \quad b_{5} = -2\mu' (\alpha'/\beta')^{2} \cos 2f' / \sin 2e', \quad b_{6} = -2\mu' \cot f' \quad (37)$$

Putting these values in eqn. (31a, b, c, d), we get results equivalent to the corresponding results given by Ben-Menahem and Singh⁶ (eqn. (3.54) and (3.56)) for isotropic media.

Discussion and Conclusions

The reflection and transmission coefficients given by Chattopadhyay and Saha⁴ for qP waves incident at the plane boundary between two monoclinic elastic half-spaces are incorrect because of two erroneous assumptions made by these authors, namely, qP waves are longitudinal (qSV waves are transverse) and the angle of reflection of qP (qSV) waves

is equal to the angle of incidence of qP (qSV) waves. In the present study, we have obtained the correct reflection and transmission coefficients by solving the problem *ab initio*.

Eqn. (32) and (33) give the amplitude ratios when plane qP waves are incident at the plane boundary between two monoclinic elastic half-spaces. In these equations, A_i/A_1 are the amplitude ratios for the horizontal component of the displacement and B_i/B_1 are the amplitude ratios for the vertical component of the displacement. Similarly, eqn. (34) and (35) give the amplitude ratios for incident qSV waves. From eqn. (17) and (21), we note that, for example, the total displacement of the incident qP waves is

$$\left(A_1^2 + B_1^2\right)^{1/2} e^{iP_1} = \left(1 + F_1^2\right)^{1/2} B_1 e^{iP_1}.$$

Therefore, the reflection coefficients can be expressed in the form

$$R_{PP} = \left(\frac{1 + F_3^2}{1 + F_1^2}\right)^{1/2} \cdot \frac{B_3}{B_1}, \quad R_{PS} = \left(\frac{1 + F_4^2}{1 + F_1^2}\right)^{1/2} \cdot \frac{B_4}{B_1}$$
(38)

for incident qP waves, and

$$R_{SP} = \left(\frac{1 + F_3^2}{1 + F_2^2}\right)^{1/2} \cdot \frac{B_3}{B_2}, R_{SS} = \left(\frac{1 + F_4^2}{1 + F_2^2}\right)^{1/2} \cdot \frac{B_4}{B_2}$$
(39)

for incident qSV waves. Similar expressions can be written for the transmission coefficients. The reflection and transmission coefficients are in terms of the six angles e_i and the six velocities $c_i(e_i)$, i = 1, 2,, 6. For an incident qP wave, e_1 and, therefore, $c_1(e_1)$ is supposed to be known. One has to compute e_i (i = 3, 4, 5, 6) for given e_1 . The velocities $c_i(e_i)$ can then be computed from explicit algebraic formulae. We give below the procedure for computing e_i for given e_1 in the case of incident qP waves and for given e_2 in the case of incident qSV waves.

The Snell's law for a monoclinic medium is given by eqn. (30) in which the apparent velocity c_a can be written as $c_a = c/p_2$, where $p(0, p_2, p_3)$ is the propagation vector. We define dimensionless apparent velocity c through the relation

$$\overline{c} = c_d/\beta = c/(p_2\beta), \tag{40}$$

where $\beta = (c_{44} / \rho)^{1/2}$. Eqn. (13) then becomes

$$\overline{c}^4 - (\overline{U} + \overline{Z}) \ \overline{c}^2 + (\overline{U} \overline{Z} - \overline{V}^2) = 0, \tag{41}$$

where

$$\overline{U} = U/(c_{44}p_2^2) = p^2 + 2\overline{c}_{24}p + \overline{c}_{22},$$

$$\overline{V} = V/(c_{44}p_2^2) = \overline{c}_{34}p^2 + (1 + \overline{c}_{23})p + \overline{c}_{24},$$

$$\overline{Z} = Z/(c_{44}p_2^2) = \overline{c}_{33}p^2 + 2\overline{c}_{33}p + 1,$$

$$p = p_3/p_2, \quad \overline{c}_{ii} = c_{ii}/c_{44}.$$
(42)

For incident qP waves, $p=-\cot e_1$; for incident qSV waves, $p=-\cot e_2$; for reflected qP waves, $p=\cot e_3$; for reflected qSV waves, $p=\cot e_4$; for transmitted qP waves, $p=-\cot e_5$; for transmitted qSV waves, $p=-\cot e_6$. For a given p, eqn. (41) can be solved for \overline{c}^2 , the two roots corresponding to qP and qSV waves. However, for a given \overline{c} , eqn. (41) is a bi-quadratic in p, corresponding to incident qP, incident qSV, reflected qP and reflected qSV. The positive roots corresponding to the reflected waves and the negative roots corresponding to the incident waves. On inserting the expressions for \overline{U} , \overline{Z} and \overline{V} from eqn. (42) into eqn. (41), the bi-quadratic in p becomes

$$g_0 p^4 + g_1 p^3 + g_2 p^2 + g_3 p + g_4 = 0, (43)$$

where $g_0 = \overline{c}_{33} - \overline{c}_{34}^2$,

$$g_1 = 2(\overline{c}_{24} \overline{c}_{33} - \overline{c}_{23} \overline{c}_{34}),$$

$$g_{2} = 1 + \overline{c}_{22} \ \overline{c}_{33} + 2 \overline{c}_{24} \ \overline{c}_{34} - (1 + \overline{c}_{23})^{2} - (1 + \overline{c}_{33}) \ \overline{c}^{2},$$

$$g_{3} = 2[\overline{c}_{22} \ \overline{c}_{34} - \overline{c}_{23} \ \overline{c}_{24} - (\overline{c}_{24} + \overline{c}_{34}) \ \overline{c}^{2}],$$

$$g_{4} = \overline{c}^{4} - (1 + \overline{c}_{22}) \ \overline{c}^{2} + \overline{c}_{22} - \overline{c}_{24}^{2}.$$

$$(44)$$

If we define $q = 1/p = p_2/p_3$, the bi-quadratic transforms to

$$g_4 q^4 + g_3 q^3 + g_2 q^2 + g_1 q + g_0 = 0. (45)$$

For angles of incidence, for which both reflected qP and reflected qSV waves exist, eqn. (45) will possess two positive roots, the smaller positive root (say q_4) corresponding to reflected SV and the larger positive root (q_3) corresponding to reflected qP. Further,

$$e_3 = \tan^{-1}(q_3), e_4 = \tan^{-1}(q_4).$$
 (46)

A similar procedure can be set up for finding e_5 and e_6 .

Acknowledgements

The authors are grateful to the Council of Scientific and Industrial Research, New Delhi for financial support through the Emeritus Scientist Scheme awarded to SJS and to the University Grants Commission, New Delhi for Junior Research Fellowship awarded to SK.

References

- 1. Musgrave, M.J.P. (1960) Geophys. J. Roy. Astr. Soc. 3: 406.
- 2. Daley, P.F. & Hron, F. (1979) Bull. Seism. Soc. Am. 67:661.
- 3. Keith, C.M. & Crampin, S. (1977) Geophys. J. Roy, Astr. Soc. 49: 181.
- 4. Chattopadhyay, A. & Saha, S. (1996) Int. J. Engng. Sci. 34: 1271.
- 5. Singh, S.J. (1999) Int. J. Engng. Sci. 37: 407.
- 6. Ben-Menahem, A. & Singh, S.J. (1981) Seismic Waves and Sources. Springer-Verlag, New York.

Appendix A

$$\Delta = \begin{vmatrix} 1 & 1 & -1 & -1 \\ F_3 & F_4 & -F_5 & -F_6 \\ a_3 & a_4 & -a_5 & -a_6 \\ b_3 & b_4 & -b_5 & -b_6 \end{vmatrix}$$

$$\Delta_3^P = \begin{vmatrix} -1 & 1 & -1 & -1 \\ -F_1 & F_4 & -F_5 & -F_6 \\ -a_1 & a_4 & -a_5 & -a_6 \\ -b_1 & b_4 & -b_5 & -b_6 \end{vmatrix}$$

$$\Delta_4^P = \begin{vmatrix} 1 & -1 & -1 & -1 \\ F_3 & -F_1 & -F_5 & -F_6 \\ a_3 & -a_1 & -a_5 & -a_6 \\ b_3 & -b_1 & -b_5 & -b_6 \end{vmatrix}$$

$$\Delta_5^P = \begin{vmatrix} 1 & 1 & -1 & -1 \\ F_3 & F_4 & -F_1 & -F_6 \\ a_3 & a_4 & -a_1 & -a_6 \\ b_3 & b_4 & -b_1 & -b_6 \end{vmatrix}$$

$$\Delta_6^P = \begin{vmatrix} 1 & 1 & -1 & -1 \\ F_3 & F_4 & -F_5 & -F_1 \\ a_3 & a_4 & -a_5 & -a_1 \\ b_3 & b_4 & -b_5 & -b_1 \end{vmatrix}$$

 Δ_3^S is obtained from Δ_3^P on replacing the elements $\{-1, -F_1, -a_1, -b_1\}$ in the first column by the elements $\{-1, -F_2, -a_2, -b_2\}$. Δ_1^S (i = 4, 5, 6) are similarly defined.

Certain generalized subclass of analytic and multivalent functions in terms of fractional calculus

P.K. BANERJI and GAMAL M. SHENAN

Department of Mathematics, J.N.V. University, Jodhpur-342 005, India.

Received April 1, 2000; Accepted August 29, 2000

Abstract

Certain generalized subclass of analytic and multivalent functions, with negative coefficients, defined in terms of certain fractional derivative, has been introduced. Present paper investigates results pertinent to coefficient estimate, distortion and inclusion theorems for the said subclass. The class, preserving integral operator of fractional order of certain kind for the generalized subclass, has been investigated too. At the conclusion of this paper, radii of *p*-valently starlikely and convexity for the said class have been determined.

(Keywords and Phrases: univalent and p-valent functions/fractional differential and integral operators/convex and starlike convex functions)

Introduction

Let A(p) denotes the class of functions of the form

$$f(z) = z^{p} + \sum_{k=n+p}^{\infty} a_{k} z^{k}; \ n = 1, 2, ...$$
 (1)

where p is a positive integer and f(z) is analytic and p-valent in the unit disc $\mathcal{D} = |z| |z| < 1$.

Also, let T(p) be a sub-class of A(p), consisting of analytic and p-valent functions, expressed in the form

$$f(z) = z^{p} - \sum_{k=n+p}^{\infty} a_{k} z^{k}; \ a_{k} \ge 0$$
 (2)

The function $f(z) \in T(p)$ is said to be in the subclass $S^*(p, \delta)$, iff

Re
$$\left\{\frac{zf'(z)}{f(z)}\right\} > \delta$$
, $z \in \mathcal{D}$, (3)

for some δ ($0 \le \delta < p$). Function $f(z) \in S^*(p, \delta)$

$$\operatorname{Re}\left\{1 + \frac{zf''(z)}{f'(z)}\right\} > \delta , \quad z \in \mathcal{D}$$
(4)

for some δ ($0 \le \delta < p$). Function $f(z) \in C(p, \delta)$ is called *p*-valent convex function of order δ . The above said classes $S^*(p, \delta)$ and $C(p, \delta)$ were introduced by Owa¹ and studied by Srivastava and Owa².

In what follows, the definition and terminologies of fractional calculus, employed in the present investigations, are due to Owa^{3,4}.

Definition 1.1: The fractional integral operator of order λ , for a function f(z), is defined by

$$D_z^{-\lambda} f(z) = \frac{1}{\Gamma(\lambda)} \int_0^z \frac{f(t)}{(z-t)^{1-\lambda}} dt , \quad \lambda > 0$$
 (5)

where f(z) is the analytic function in a simply-connected region of the z-plane, containing the origin, and the multiplicity of $(z-t)^{\lambda-1}$ is removed by requiring $\log(z-t)$ to be real, when (z-t) > 0.

Definition 1.2: Fractional derivative of order λ , for a function f(z), is defined as

$$D_z^{\lambda} f(z) = \frac{1}{\Gamma(1-\lambda)} \frac{d}{dz} \int_0^z \frac{f(t)}{(z-t)^{\lambda}} dt , \quad 0 \le \lambda < 1$$
 (6)

while the definition of f(z) and its validity is given in (5) and multiplicity of $(z-t)^{\lambda-1}$ is removed as Definition 1.1.

Definition 1.3: Under the hypothesis of definition 1.2, the fractional derivative of order $(n + \lambda)$, for f(z), is expressed as

$$D_z^{n+\lambda} f(z) = \frac{d^n}{dz^n} D_z^{\lambda} f(z) , \quad 0 \le \lambda < 1$$
 (7)

where $n \in N_0 = N \cup \{0\}$, where N is the set of natural numbers.

Let us define the following fractional differential operator $U_z^{(\lambda,\,p)}$, $0\leq \lambda <1$.

Definition 1.4: For a real number λ ($0 \le \lambda < 1$), we define

$$U_z^{(\lambda, p)}: T(p) \to T(p)$$

by

$$U_z^{(\lambda, p)} f(z) = z^p - \sum_{k=n+p}^{\infty} a_k G_p(k, \lambda) z^k,$$
(8)

where

$$G_{p}(k,\lambda) = \frac{\Gamma(k+1)\Gamma(p+1-\lambda)}{\Gamma(p+1)\Gamma(k+1-\lambda)}.$$
(9)

For $z \neq 0$, we define

$$U_z^{(\lambda, p)} f(z) = \frac{\Gamma(p+1-\lambda)}{\Gamma(p+1)} z^{\lambda} D_z^{\lambda} f(z), \qquad (10)$$

where $D_z^{\lambda} f(z)$ is the fractional derivative of f, of order λ ($0 \le \lambda < 1$), while

$$U_z^{(0,p)} f(z) = f(z) ; U_z^{(1,p)} f(z) = \frac{z f'(z)}{p}.$$
 (11)

Now, with the help of operator $U_z^{\lambda, p} f(z)$, we may introduce following class of functions in T(p).

Definition 1.5: A function $f(z) \in T(p)$ is said to be in the class $S_p^*(A, B, \alpha, \beta, \lambda)$, iff

$$\left| \left\{ \frac{z \left(U_z^{(\lambda, p)} f(z) \right)'}{U_z^{(\lambda, p)} f(z)} - p \right\} / \left\{ (p - \alpha) \left(A - B \right) + pB - Bz \frac{\left(U_z^{(\lambda, p)} f(z) \right)'}{U_z^{(\lambda, p)} f(z)} \right\} \right| < \beta$$
 (12)

where $z \in \mathcal{D}$ and it validity is given by $-1 \le B < A < 1$, $0 < A \le 1$, $0 < \beta \le 1$, $0 \le \alpha < p$ and $0 \le \lambda < 1$.

The sub-class $S_p^*(A, B, \alpha, \beta, \lambda)$, defined by (12) is much general in character and as such number of subsclasses become apparent cases of it, which were studied by various authors, we mention some of them

(i)
$$S_p^*(1,-1,\alpha,1,0) = S^*(p,\alpha)$$
 ; (cf. 1)

(ii)
$$S_p^*(1,-1,\alpha,1,1) = C(p,\alpha)$$
 ; (cf. 1)

(iii)
$$S_1^*(1,-1,\alpha,1,\lambda) = S_{\lambda}(\alpha)$$
 ; (cf. 5)

(iv)
$$S_1^*(1,-1,\alpha,\beta,0) = S(\alpha,\beta)$$
 ; (cf. 6)

(v)
$$S_1^*(1,-1,\alpha,\beta,1) = C(\alpha,\beta)$$
 ; (cf. 6)

(vi)
$$S_p^*(1, 1-2\zeta, \alpha, \beta, 0) = S_p^*(\alpha, \beta, \zeta)$$
 ; (cf. 3); $1/2 < \zeta \le 1$.

(vii)
$$S_p^*(A, B, \alpha, 1, 0) = T_p^*(A, B, \alpha)$$
 ; (cf. 8)

(viii)
$$S_p^*(A, B, \alpha, 1, 1) = C_p(A, B, \alpha)$$
; (cf. 8).

Several other such subclass may be formulated by specializing parameters.

Coefficients Estimates

Theorem 1: A function f(z), defined by (2), is in the class $S_p^*(A, B, \alpha, \beta, \lambda)$, iff

$$\sum_{k=n+p}^{\infty} \left[(1 - \beta B) (k - p) + \beta (A - B) (p - \alpha) \right] G_p(k, \lambda) a_k \le (A - B) \beta (p - \alpha), \quad (13)$$

where $G_p(k, \lambda)$ is given in (9).

The inequality (13) is sharp.

Proof: Assume that (13) holds true, and let |z| = 1, then we have

$$\left| z \left(U_{z}^{(\lambda, p)} f(z) \right)' - p \ U_{z}^{(\lambda, p)} f(z) \right| - \beta \left| (p - \alpha) (A - B) + p B \ U_{z}^{(\lambda, p)} f(z) - B z \left(U_{z}^{(\lambda, p)} f(z) \right)' \right|$$

$$= \left| -\sum_{k=n+p}^{\infty} (k - p) \ G_{p} (k, \lambda) \ a_{k} z^{k} \right|$$

$$- \beta \left| (A - B) (p - \alpha) z^{p} + \sum_{k=n+p}^{\infty} \left| B (k - p) + \beta (A - B) (p - \alpha) \right| G_{p} (k, \lambda) \ a_{k} z^{k} \right|$$

$$\leq \sum_{k=n+p}^{\infty} \left| (1 - B\beta) k - (1 - A\beta) p - \beta (A - B) \alpha \right| G_{p} (k, \lambda) \ a_{k} - (A - B) \beta (p - \alpha) \leq 0.$$

Hence, by the principle of maximum modulus, it is asserted that $f(z) \in S_p^*(A, B, \alpha, \beta, \lambda)$.

Conversely, we suppose that f(z) is in the class S_p^* $(A, B, \alpha, \beta, \lambda)$, then by virtue of (12)

$$\left| \left\{ \frac{z \left(U_z^{(\lambda,p)} f(z) \right)'}{U_z^{(\lambda,p)} f(z)} - p \right\} / \left\{ (p-\alpha) \left(A - B \right) + pB - Bz \frac{\left(U_z^{(\lambda,p)} f(z) \right)'}{U_z^{(\lambda,p)} f(z)} \right\} \right|$$

$$= \frac{-\sum_{k=n+p}^{\infty} (k-p) G_{p}(k,\lambda) a_{k} z^{k}}{(A-B) (p-\alpha) z^{p} + \sum_{k=n+p}^{\infty} |B(k-p) - (A-B) (p-\alpha)| G_{p}(k,\lambda) a_{k} z^{k}} < \beta.$$
 (14)

Since $|Re(z)| \le |z|$ for any z, we find from (14), that,

$$\operatorname{Re}\left[\frac{\displaystyle\sum_{k=n+p}^{\infty}\left(k-p\right)\,G_{p}\left(k,\lambda\right)\,a_{k}\,z^{k}}{\left(A-B\right)\left(p-\alpha\right)\,z^{p}+\displaystyle\sum_{k=n+p}^{\infty}\left|B\left(k-p\right)-\left(A-B\right)\left(p-\alpha\right)\right|\,G_{p}\left(k,\lambda\right)\,a_{k}\,z^{k}}\right]<\beta.$$

Now choose values of z on the real axis, such that

$$\frac{z \left(U_z^{(\lambda, p)} f(z)\right)'}{U_z^{(\lambda, p)} f(z)}$$

is real. Assuming $z \to 1$, through real values, we get

$$\sum_{k=n+p}^{\infty} (k-p) G_p(k,\lambda) a_k \leq (A-B) \beta(p-\alpha) + \sum_{k=n+p}^{\infty} \beta[B(k-p) - (A-B)(p-\alpha)] G_p(\lambda,p) \alpha_k,$$

which asserts (13).

Corollary 1: Let the function f(z), defined by (2), belongs to the class S_p^* $(A, B, \alpha, \beta, \lambda)$, then

$$a_k \le \frac{(A-B)\beta(p-\alpha)}{\left[(1-\beta B)(k-p)+\beta(A-B)(p-\alpha)\right]G_p(k,\lambda)}, \ k \ge n+p. \tag{15}$$

Remark 1: Considering $\lambda = 0$ and $\lambda = 1$ respectively in Theorem 1, results due to Singh and Sohi⁸ become particular cases of our investigation.

Similarly, by setting $p = A = \beta = -B = 1$ in Theorem 1, result of Srivastava, Mishra and Das⁵ become particular cases of our result.

Distortion Theorem

Theorem 2: If f(z), defined by (2), belongs to the class $S_p^*(A, B, \alpha, \beta, \lambda)$, $0 \le \lambda < 1$; then for $0 \le \mu < 1$, we have

$$|U_{z}^{(\mu,p)}f(z)| \ge |z|^{p} - \frac{(p+1)_{n}(A-B)\beta(p-\alpha)}{(p+1-\mu)_{n}G_{p}(n+p,\lambda)|(1-\beta B)n+\beta(A-B)(p-\alpha)|} |z|^{n+p}$$
 (16)

and

$$|U_{z}^{(\mu,p)}f(z)| \le |z|^{p} + \frac{(p+1)_{n}(A-B)\beta(p-\alpha)}{(p+1-\mu)_{n}G_{p}(n+p,\lambda)|(1-\beta B)n+\beta(A-B)(p-\alpha)|}|z|^{n+p}$$
(17)

for $z \in \mathcal{D}$. The estimates are sharp.

Proof: For $0 \le \lambda < 1$,

$$\left\{\frac{\Gamma(k+1)}{\Gamma(k+1-\lambda)}\right\}_{k=n+p}^{\infty}$$

is an increasing sequence, and thus, from Theorem 1, we have

$$G_p(n+p,\lambda)$$
 $\sum_{k=n+p}^{\infty} \left\{ (1-\beta B) (k-p) + \beta (A-B) (p-\alpha) \right\} a_k$

$$\leq \sum_{k=n+p}^{\infty} \left\{ (1-\beta B) \left(k-p \right) + \beta \left(A-B \right) \left(p-\alpha \right) \right\} G_p \left(k,\lambda \right) a_k \leq (A-B) \beta \left(p-\alpha \right)$$

i.e.
$$\sum_{k=n+p}^{\infty} \left| (1 - \beta B) (k - p) + \beta (A - B) (p - \alpha) \right| a_k \le \frac{(A - B) \beta (p - \alpha)}{G_p (n + p, \lambda)}, \tag{18}$$

which yields

$$\sum_{k=n+p}^{\infty} a_k \le \frac{(A-B)\beta(p-\alpha)}{G_p(n+p,\lambda)\left[(1-\beta B)n+(A-B)(p-\alpha)\right]}, \ k \ge n+p.$$
 (19)

Now, due to mutual adjustments of terms in (18) and (19), we may write the following:

$$\sum_{k=n+p}^{\infty} (1-B\beta) k a_k \leq \frac{(A-B)\beta(p-\alpha)}{G_p(n+p,\lambda)} \left| (1-\beta B)p - \beta(A-B)(p-\alpha) \right| \sum_{k=n+p}^{\infty} a_k.$$

$$\leq \frac{(A-B)\beta(p-\alpha)}{G_p(n+p,\lambda)} + \frac{\left| (1-\beta B)p - \beta(A-B)(p-\alpha) \right| (A-B)\beta(p-\alpha)}{G_p(n+p,\lambda) \left| (1-\beta B)n + \beta(A-B)(p-\alpha) \right|}$$

$$= \frac{(A-B)\beta(p-\alpha)(1-\beta B)(n+p)}{G_p(n+p,\lambda) \left| (1-\beta B)n + \beta(A-B)(p-\alpha) \right|}$$

i.e.
$$\sum_{k=n+p}^{\infty} k \, a_k = \frac{(A-B) \, \beta \, (p-\alpha) \, (n+p)}{G_p \, (n+p, \, \lambda) \, |(1-\beta B) \, n+\beta \, (A-B) \, (p-\alpha)|} \,. \tag{20}$$

From (8), we have the expression, as follows:

$$U_z^{(\mu,p)} f(z) = z^p - \sum_{k=n+p}^{\infty} l_k z^k,$$
 (21)

where

$$l_k = \frac{\Gamma(k+1)\Gamma(p+1-\mu)}{\Gamma(p+1)\Gamma(k+1-\mu)} a_k , \quad k \ge n+p.$$
 (22)

We observe that, for $0 \le \mu < 1$,

$$l \leq \frac{\Gamma(k+1)\Gamma(p+1-\mu)}{\Gamma(p+1)\Gamma(k+1-\mu)}$$

$$= \frac{\Gamma(p+1-\mu)}{\Gamma(k+1-\mu)} \frac{k}{(k-\mu)} \cdot \frac{(k-1)}{(k-\mu-1)} \cdots \frac{(1-n+p)}{(n+p-\mu+1)} \cdot \frac{\Gamma(n+p+1)}{\Gamma(n+p-\mu+1)}$$

$$\leq \frac{\Gamma(p+1-\mu)}{\Gamma(k+1-\mu)} \frac{k}{k-1} \cdot \frac{(k-1)}{(k-2)} \cdots \frac{(n+p+1)}{(n+p)} \cdot \frac{\Gamma(n+p+1)}{\Gamma(n+p-\mu+1)}$$

$$= \frac{\Gamma(p+1-\mu)\Gamma(n+p)}{\Gamma(p+1)\Gamma(n+p-\mu+1)} k.$$
(23)

Now a mutual employment of (20) and (23), allows us to write

$$\sum_{k=n+p}^{\infty} l_k \le \frac{\Gamma(p+1-\mu)\Gamma(n+p)}{\Gamma(p+1)\Gamma(n+p-\mu+1)} \sum_{k=n+p}^{\infty} k \, a_k$$

$$= \frac{(p+1)_n (A-B) \, \beta(p-\alpha)}{(p+1-\mu)_n \, G_n(n+p,\lambda) \, |(1-\beta B) \, n+\beta \, (A-B) \, (p-\alpha)|} \, . \tag{24}$$

Eventually

$$|U_{z}^{(\mu, p)} f(z)| \ge |z|^{p} - |z|^{n+p} \sum_{k=n+p}^{\infty} l_{k}$$

yields the results (16) due to (24). Similarly, (17) can be proved.

The sharpness of the theorem follows, if we consider

$$f(z) = z^{p} - \frac{(A-B)\beta(p-\alpha)}{G_{p}(n+p,\lambda)\left[(1-\beta B)n+\beta(A-B)(p-\alpha)\right]}.$$
 (25)

Corollary 2: By setting $\mu = 0$ in Theorem 2, we obtain, for $f(z) \in S_p^* (A, B, \alpha, \beta, \lambda)$, the following:

$$|f(z)| \ge |z|^{p} - \frac{(A-B)\beta(p-\alpha)}{G_{p}(n+p,\lambda)|(1-\beta B)n+\beta(A-B)(p-\alpha)|} |z|^{n+p}$$
 (26)

and

$$|f(z)| \le |z|^p + \frac{(A-B)\beta(p-\alpha)}{G_n(n+p,\lambda)|(1-\beta B)n+\beta(A-B)(p-\alpha)|} |z|^{n+p}.$$
 (27)

Estimates are, thus, sharp for f(z) given by (3.10).

Corollary 3: Assuming $\mu \to 1$ and for $f(z) \in S_p^*(A, B, \alpha, \beta, \lambda)$, Theorem 2 reduces to

$$|f'(z)| \ge p |z|^p - \frac{(A-B)\beta(p-\alpha)(p+n)}{G_n(n+p,\lambda)|(1-\beta B)n + \beta(A-B)(p-\alpha)|} |z|^{n+p}, \qquad (28)$$

and

$$|f'(z)| \le p |z|^{p-1} - \frac{(A-B)\beta(p-\alpha)(p+n)}{G_p(n+p,\lambda)|(1-\beta B)n+\beta(A-B)(p-\alpha)|} |z|^{n+p-1}.$$
 (29)

Sharpness of the estimation is verified as Cor. 3.

Inclusion Theorem

Theorem 3: For $0 \le \lambda \le \mu < 1$, $0 \le \alpha < p$, $0 < \beta \le 1$, $-1 \le B < A \le 1$ and $0 < A \le 1$, we have

$$S_p^*(A, B, \alpha, \beta, \mu) \subseteq S_p^*(A, B, \alpha, \beta, \lambda).$$
(30)

Proof: For $f(z) \in S_p^*(A, B, \alpha, \beta, \mu)$, Theorem 1 reduces to

$$\sum_{k=n+p}^{\infty} \left| (1 - \beta B) (k - p) + \beta (A - B) (p - \alpha) \right| G_p(k, \mu) a_k \le (A - B) \beta (p - \alpha), \quad (31)$$

where, $G_p(k, \mu)$ may be expressed, by virtue of (9), as

$$G_{p}\left(k,\,\mu\right)\,=\,\frac{\Gamma\left(k+\,1\right)\,\Gamma\left(p+\,1\,-\,\mu\right)}{\Gamma\left(p+\,1\right)\,\Gamma\left(k+\,1\,-\,\mu\right)}\,.$$

For μ to be resitricted as above, and ϕ (μ) be assumed as an increasing function

$$\phi(\mu) = \frac{\Gamma(p+1-\mu)}{\Gamma(k+1-\mu)}$$

we observe that

$$\phi(\mu+1)-\phi(\mu)=\frac{\Gamma(p-\mu)}{\Gamma(k+1-\mu)}(k-p)>0,$$

very clearly asserts that $\phi(\mu)$ is definitely an increasing function of λ [since (k-p)>0] and thus,

$$G_p(k,\lambda) \le G_p(k,\mu). \tag{32}$$

Finally from (31), we have

$$\sum_{k=n+p}^{\infty} \left| (1 - \beta B) (k - p) + \beta (A - B) (p - \alpha) \right| G_p(k, \lambda) a_k$$

$$\leq \sum_{k=n+p}^{\infty} \left| (1 - \beta B) (k - p) + \beta (A - B) (p - \alpha) \right| G_p(k, \mu) a_k$$

$$\leq (A - B) \beta (p - \alpha).$$

This is an evidence, that, $f(z) \in S_p^*(A, B, \alpha, \beta, \lambda)$.

Fractional Integral Operator

Theorem 4: Let $f(z) \in S_p^*(A, B, \alpha, \beta, \lambda)$, be defined by (2) and c be real number, such that c > -p, then the function F(z), define by

$$F(z)' = \frac{c+p}{z^c} \int_0^z t^{c-1} f(t) dt,$$
 (33)

also belongs to the above class.

Proof: By virtue of (33), it follows from (2) that,

$$F(z) = \frac{c+p}{z^c} \int_0^z t^{c-1} \left\{ t^p - \sum_{k=n+p}^\infty a_k t^k \right\} dt,$$

which upon integration and change of order of summation and integration, permissible under stated conditions, reduces to

$$F(z) = z^{p} - \sum_{k=n+p}^{\infty} b_{k} z^{k},$$
 (34)

where

$$b_k = \left(\frac{c+p}{c+k}\right) a_k. \tag{35}$$

Thus, we write

$$\sum_{k=n+p}^{\infty} \left| (1 - \beta B) (k - p) + \beta (A - B) (p - \alpha) \right| G_p(k, \lambda) b_k$$

$$= \sum_{k=n+p}^{\infty} \left| (1 - \beta B) (k - p) + \beta (A - B) (p - \alpha) \right| G_p(k, \lambda) \left(\frac{c + p}{c + k} \right) a_k$$

$$\leq (A - B) \beta (p - \alpha).$$

which is the justification, by virtue of Theorem 1, for $F(z) \in S_p^*(A, B, \alpha, \beta, \lambda)$, which completes the proof.

Theorem 5: Let c be a real number, such that c < -p, if $F(z) \in S_p^*$ $(A, B, \alpha, \beta, \lambda)$, then the function f(z), which is defined by (33), is p-valent in $|z| < R_1$, where

$$R_{1} = \inf_{k \ge n+p} \left[\frac{(c+p) \Gamma(k) \Gamma(p+1-\lambda) \left| (1-\beta B) (k-p) + \beta (A-B) (p-\alpha) \right|}{(c+k) \Gamma(p) \Gamma(k+1-\lambda) (A-B) \beta (p-\alpha)} \right]^{1/(k-p)}$$
(36)

and the result is sharp.

Proof: Let

$$F(z) = z^p - \sum_{k=n+p}^{\infty} a_k z^k,$$

it follows from (33), that,

$$f(z) = \frac{z^{1-c} \left(z^c F(z)\right)'}{c+p} ,$$

i.e.
$$f(z) = z^p - \sum_{k=n+p}^{\infty} \left(\frac{c+k}{c+p}\right) a_k z^k, \ c > -p.$$
 (37)

To prove the required result, it is sufficient to prove that

$$\left| \frac{f'(z)}{z^{p-1}} - p \right| \le p \,, \quad |z| < R_1 \,. \tag{38}$$

Now,

$$\left|\frac{f'(z)}{z^{p-1}}-p\right| = \left|-\sum_{k=n+p}^{\infty} k\left(\frac{c+k}{c+p}\right) a_k z^{k-p}\right|,$$

thus, if

$$\left|\frac{f'(z)}{z^{p-1}}-p\right| \leq p,$$

then

$$\sum_{k=n+p}^{\infty} \frac{k}{p} \left(\frac{c+k}{c+p} \right) a_k \mid z \mid^{k-p} \le 1.$$
 (39)

Also, since $F(z) \in S_p^*(A, B, \alpha, \beta, \lambda)$, so by Theorem 1, we have

$$\sum_{k=n+p}^{\infty} \frac{\left| (1-\beta B) (k-p) + \beta (A-B) (p-\alpha) \right| G_p(k,\lambda) a_k}{(A-B) \beta (p-\alpha)} \le 1. \tag{40}$$

Hence (39) will be satisfied, if

$$\frac{k}{p}\left(\frac{c+k}{c+p}\right) \left|z\right|^{k-p} \leq \frac{\left|(1-\beta B)\left(k-p\right)+\beta\left(A-B\right)\left(p-\alpha\right)\right| G_{p}\left(k,\lambda\right)}{(A-B)\beta\left(p-\alpha\right)}$$

i.e.
$$|z| \le \left\lceil \frac{(c+p)\Gamma(k)\Gamma(p+1-\lambda)\left|(1-\beta B)\left(k-p\right)+\beta(A-B)\left(p-\alpha\right)\right|}{(c+k)\Gamma(p)\Gamma(k+1-\lambda)\left(A-B\right)\beta\left(p-\alpha\right)} \right\rceil^{1/(k-p)}, k \ge n+p.$$
 (41)

Therefore, this proves that f(z) is p-valent in $|z| < R_1$. Sharpness follows, if we consider

$$f(z) = z^{p} - \frac{(c+k)(A-B)\beta(p-\alpha)}{(c+p)(1-\beta B)(k-p) + \beta(A-B)(p-\alpha)|G_{n}(k,\lambda)}, k \ge n+p. \quad (42)$$

Radii of Starlikeness and Convexity

Theorem 6: The function f(z), defined by (2) be in the class $S_p*(A, B, \alpha, \beta \lambda)$, then $f(z) \in S^*(p, \delta)$, $0 \le \delta < p$ in $|z| < R_2$, where

$$R_{2} = \inf_{k \ge n+p} \left[\frac{(p-\delta) \left| (1-\beta B) (k-p) + \beta (A-B) (p-\alpha) \right|}{(k-\delta) (A-B) \beta (p-\alpha)} \right]^{1/(k-p)}$$
(43)

The result is sharp.

Proof: In order to establish the required result, we deem it sufficient to prove that

$$\left|\frac{zf'(z)}{f(z)} - p\right| \le p - \delta, \quad |z| \le R_2. \tag{44}$$

Now

$$\left|\frac{zf'(z)}{f(z)} - p\right| = \left| -\frac{\sum\limits_{k=n+p}^{\infty} (k-p) \, a_k \, z^k}{z^p - \sum\limits_{k=n+p}^{\infty} a_k \, z^k} \right|.$$

$$\leq \frac{\sum_{k=n+p}^{\infty} (k-p) a_k |z|^{k-p}}{1 - \sum_{k=n+p}^{\infty} a_k |z|^{k-p}},$$

thus if,

$$\left|\frac{zf'(z)}{f(z)} - p\right| \le p - \delta ,$$

then, it is obvious to write

$$\sum_{k=n+p}^{\infty} \left(\frac{k-\delta}{p-\delta} \right) .a_k \, z^{k-p} \le 1 \, . \tag{45}$$

Also, as $f(z) \in S_p^*(A, B, \alpha, \beta, \lambda)$, by virtue of Theorem 1, we write

$$\sum_{k=n+p}^{\infty} \frac{\left| (1-\beta B) (k-p) + \beta (A-B) (p-\alpha) \right| G_p(k,\lambda)}{(A-B) \beta (p-\alpha)} \leq 1.$$

Therefore, (45) is justified, if

$$|z| \le \left\lceil \frac{(p-\delta) \left| (1-\beta B) (k-p) + \beta (A-B) (p-\alpha) \right|}{(k-\delta) (A-B) \beta (p-\alpha)} G_p(k,\lambda) \right\rceil^{1/(k-p)}. \tag{46}$$

This concludes that, $f(z) \in S^*(p, \delta)$, i.e. f(z) is p-valently starlike of order δ in $|z| < R_2$. Sharpness follows from the function defined as

$$f(z) = z^{p} - \frac{(A - B) \beta (p - \alpha)}{|(1 - \beta B) (k - p) + \beta (A - B) (p - \alpha)| G_{p}(k, \lambda)} z^{k}.$$
 (47)

Theorem 7: Let the function f(z), defined by (2), belongs to $S_p^*(A, B, \alpha, \beta, \lambda)$, then $f(z) \in C(p, \delta)$, $0 \le \delta < p$ in the disc $|z| < R_3$, where

$$R_{3} = \inf_{k \ge n+p} \left[\frac{p(p-\delta) \left| (1-\beta B)(k-p) + \beta (A-B)(p-\alpha) \right|}{k(k-\delta)(A-B)\beta(p-\alpha)} G_{p}(k,\lambda) \right]^{1/(k-p)}$$
(48)

The result is sharp.

Proof: The prove the statement completely, it is sufficient to show that

$$\left|\left\{1 \ + \frac{zf''(z)}{f'(z)}\right\} \ - \ p\right| \ \leq \ (p-\delta) \ , \ \ |z| \ \leq \ R_3 \ .$$

we have,

$$\left| \left\{ 1 + \frac{zf''(z)}{f'(z)} \right\} - p \right| = \begin{vmatrix} \sum_{k=n+p}^{\infty} k (k-p) a_k z^{k-p} \\ -\sum_{k=n+p}^{\infty} k a_k z^{k-p} \end{vmatrix}$$

$$\sum_{k=n+p}^{\infty} k (k-p) a_k |z|^{k-p} \\
\leq \frac{\sum_{k=n+p}^{\infty} k a_k |z|^{k-p}}{p - \sum_{k=n+p}^{\infty} k a_k |z|^{k-p}}.$$
(50)

Therefore, if (49) is true, then

$$\sum_{k=n+p}^{\infty} k (k-p) a_k |z|^{k-p} \le (p-\delta) \left\{ p - \sum_{k=n+p}^{\infty} k a_k |z|^{k-p} \right\}$$

i.e.

$$\sum_{k=n+p}^{\infty} \frac{k}{p} \cdot \frac{(k-\delta)}{(p-\delta)} a_k |z|^{k-p} \le 1.$$

$$(51)$$

By virtue of Theorem 1, (13) holds true, and thus (51) will be justified if

$$\left|z\right| \leq \left[\frac{p\left(p-\delta\right)\left|\left(1-\beta B\right)\left(k-p\right)+\beta(A-B)\left(p-\alpha\right)\right|\,G_{p}\left(k,\lambda\right)}{k\left(k-\delta\right)\left(A-B\right)\,\beta\left(p-\alpha\right)}\right]^{1/(k-p)}.$$

Thus $f(z) \in C(p, \delta)$, i.e. f(z) is p-valently convex of order δ in $|z| < R_3$, while sharpness for f(z) follows from (47). The theorem is thereby asserted.

Acknowledgements

This work is supported by CSIR Research Scheme No. 25 (0105)/99/EMR-II, sanctioned to first author (PKB). The second author (GMS) is thankful to Al-Azhar University of Gaza, Palestine, for financial assitance.

References

- 1. Owa, S. (1985) Simon Stevin, 4(59): 385.
- 2. Srivastava, H.M. & Owa, S. (1987) J. Math. Anal. Appl. 122 (1): 187.
- 3. Owa, S. (1976) Kyungpook Math. J. 18:53.
- Owa, S. (1985) Some applications of fractional calculus. In 'Fractional Calculus', A.C. MacBridge and G.F. Roach (Eds.), Research Notes in Mathematics, No. 138, 164-175, Pitman Advanced Publ. Prog., Pitman Publishing Inc. Boston, London.

- 5. Srivastava, H.M., Mishra, A.K. & Das, M.K. (1998) Comm. Appl. Anal. 2 (3): 321.
- 6. Gupta, V.P. & Jain, P.K. (1976) Bull. Australian Math. Soc. 14: 409.
- 7. Kulkarni, S.R., Joshi, S.B. & Aouf, M.K. (1994) J. Ramanujan Math. Soc. 9(1): 55.
- 8. Singh, Amir & Sohi, N.S. (1995) Pure Appl. Math. Sci. 12(1-2): 53.

On the equation for electron – proton pairing at finite temperature

L.K. PANDE

Theoretical Physics Group, SES, Jawaharlal Nehru University, New Delhi-110067, India.

Received July 22, 2000; Accepted April 10, 2001

Abstract

An equation was considered recently to describe electron-proton pairing at finite temperature. It was shown to have a solution in terms of a difference equation which, on numerical evaluation, led to a temperature—dependent bound state spectrum. In the present paper, the solution of the pairing equation is discussed afresh, resulting in an independent (and elementary) derivation of the crucial difference equation.

(Keywords: finite temperature theory/electron proton pairing/x-ray astronomy)

Introduction

Some time back, an equation was obtained which described the pairing of an electron and a proton at finite temperature. Under a suitable approximation, this equation yielded a bound state spectrum in the temperature regime around one million degrees Kelvin. The deepest of the binding energies predicted in this spectrum were in the KeV range. This meant that the bound states could actually sustain themselves in the system, despite its high temperature, since the background thermal energy per particle, which would only be in the range of few hundred eV, would not be enough to break up the former.

The above finding may be quite relevant to an understanding of Solar coronal active regions and solar flares. In particular, the mechanism for the massive energy production associated with these phenomena could be provided by falling into the paired state by the electrons and the protons of the astrophysical plasma. Furthermore, the emission spectrum of the paired state in certain possible stable or quazi-stable configurations corresponding to definite temperatures, could also explain a large part of the X-ray emission line data from the flares and the other active regions of the Sun and its corona, Preliminary work along these lines reported in the reference quoted above and in later papers²⁻³ has led to encouraging results.

340 L.K. PANDE

Of course, the occurance of sustainable bound states at high temperatures is a rather unexpected result. So despite its good consequences, the basis of this result needs detailed probing. Some work in this direction has been published⁴. More recently the author⁵ has given a systematic derivation of a coupled set of equations for electron-proton pairing at finite temperature and shown that in the static Coulomb limit they indeed lead to the pairing equation used by us earlier. In the present paper, we try a new approach to solving the approximate version of the pairing equation. This is done by transforming the equation onto coordinate space where the ansatz for the solution can be taken as a sum over the familier Laguerre polynomials. The coefficients in the sum are then found to obey a difference equation the solution of which leads to the bound state spectrum. This difference equation is identical to the one obtained earlier from momentum-space methods using the less familier four-dimensional spherical harmonics given in terms of Gegenbauer polynomials.

The Pairing Equation in Coordinate Space

The pairing equation in momentum space in the static Coulomb limit is given by

$$\left(W - \frac{p^2}{2\mu}\right) \psi (\overrightarrow{p}) = \frac{\alpha}{4\pi^2} Q(w, p) \int \frac{d\overrightarrow{q}}{(\overrightarrow{q} - \overrightarrow{p})^2} \psi (\overrightarrow{q}), \qquad (1)$$

$$Q(w,p) = \tan h \left[(\mu_a w - \epsilon_p^a) \frac{\beta}{2} \right], \tan h \left[(\mu_b w - \epsilon_p^b) \frac{\beta}{2} \right]$$
 (2)

where w = -|w| stands for the pair binding energy, \overrightarrow{p} or \overrightarrow{q} for the (relative) pair momentum; $\alpha = (e^2 / 4\pi)$ is the fine structure constant in natural units $((\hbar = c = 1)$ which we use throughout; a and b stand, respectively, for the electron and the proton and

$$\mu = \frac{m_a m_b}{m_a + m_b}, \quad \mu_{a,b} = \frac{m_{a,b}}{m_a + m_b}$$
 (3)

$$\in_p^{a,b} = \frac{p^2}{2m_{a,b}}, \ \beta = \frac{1}{KT}$$

where k is the Boltzmann constant and T the temperature.

It may be noted that

$$Q(w,p) = \frac{a - \tan h (\beta p^2 / 4m_a)}{1 - a \tan h (\beta p^2 / 4m_a)} - \frac{b - \tan h (\beta p^2 / 4m_b)}{1 - b \tan h (\beta p^2 / 4m_b)}$$
(4)

where

$$a = \tan h \ (\beta \mu_a w / 2), \quad b = \tan h \ (\beta \mu_b w / 2).$$
 (5)

We now make the approximation

$$\tan h \left(\frac{\beta p^2}{4m_{a,b}} \right) = \frac{\beta p^2}{4m_{a,b}} \tag{6}$$

and combine the two terms in eqn. (4), dropping terms of order p^4 and higher in both the numerator and the denominator in keeping with the spirit of (6). So we get

$$Q(w,p) = \frac{d_1 - d_2 p^2}{1 - d_3 p^2} \equiv Q(w,p)$$
 (7)

where

$$d_1 = a + b$$
, $d_2 = \beta (1 + ab) / 4\mu$, $d_3 = \frac{\beta}{4} \left(\frac{a}{m_a} + \frac{b}{m_b} \right)$ (8)

Next, we replace Q in eqn. (1) by \overline{Q} and write the resulting approximate version of the equation as

$$(1 - d_3 p^2) \left(w - p^2 / 2\mu \right) \psi \left(\overrightarrow{p} \right) = \frac{\alpha}{4\pi^2} \left(d_1 - d_2 p^2 \right) \int \frac{d\overrightarrow{q}}{(\overrightarrow{q} - \overrightarrow{p})^2} \psi \left(\overrightarrow{q} \right)$$
(9)

On multiplying both sides by

$$\frac{1}{(2\pi)^{3/2}} \int d\overrightarrow{p} e^{-\overrightarrow{p}\cdot\overrightarrow{r}}$$

and going through the standard manipulations, we can transform this equation onto coordinate space to get

342 L.K. PANDE

$$(1+d_3\,\nabla^2)\left(W\,+\,\frac{\nabla^2}{2\,\mu}\right)\,\psi\ (\overrightarrow{r}) = \frac{\alpha}{2}\left(\frac{d_1}{r}\,\,\psi\,(\overrightarrow{r}) +\,d_2\,\nabla^2\left[\frac{\psi\,(\overrightarrow{r})}{r}\right]\right) \quad (10)$$

where

$$\Psi(\overrightarrow{r}) = \frac{1}{(2\pi)^{3/2}} \int d\overrightarrow{p} e^{-i\overrightarrow{p}\cdot\overrightarrow{r}} \Psi(\overrightarrow{p}). \tag{11}$$

Solution of the Pairing Equation in Coordinate Space

We observe that in the absence of the terms containing d_2 and d_3 , the pairing equation in coordinate space, namely, eqn. (10) above, has the same structure as the Schrodinger equation for the hydrogen atom. The solution in that case is:

$$\psi(\overrightarrow{r}) = R(r) Y_i^m(\theta, \phi) = \frac{u(r)}{r} Y_i^m(\theta, \phi)$$
 (12)

$$u(r) = u_{m,l}(r) = N' r^{l+1} e^{-p_0 r} L_{n-l-1}^{2l+1}(2p_0 r)$$
(13)

where $Y_l^m(\theta, \phi)$ are the three dimensional spherical harmonics with l and m standing, respectively, for the angular momentum and the azimuthal quantum numbers (l = 0, 1, 2,; m for a given l takes the 2l + 1 values -l, -l+1,, l-1, l); the L's denote the Laguerre polynomials in which the index n, the principal quantum number, takes on the values l+1, l+2,, l.e., for any given $l, n \ge l+1$; N' is the normalisation constant.

Now, as regards eqn. (10), we can once more try separation of variables, and choose $\psi = RY = (u/r) Y$ with unknown R (or u) and Y. The spherical symmetry of the problem enables us to choose Y_l^m for Y. So we can still use eqn. (12). We can then also write (see for instance Ch. 7 in Powell and Crasemann⁶).

$$\nabla^{2} = \frac{1}{r^{2}} \frac{\partial}{\partial r} \left(r^{2} \frac{\partial}{\partial r} \right) + \frac{1}{r^{2}} \left[\frac{1}{\sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) + \frac{1}{\sin^{2} \theta} \frac{\partial^{2}}{\partial \phi^{2}} \right]$$

$$= \frac{1}{r^{2}} \frac{\partial}{\partial r} \left(r^{2} \frac{\partial}{\partial r} \right) - \frac{l(l+1)}{r^{2}} . \tag{14}$$

With the use of eqn. (12) and (14), eqn. (10) takes the form

$$(1 + d_3 \nabla^2) \left(-p_0^2 \frac{u}{r} Y_l^m + \frac{1}{r} \frac{d^2 u}{dr^2} Y_l^m - \frac{l(l+m)}{r^2} \frac{u}{r} Y_l^m \right)$$

$$= \alpha \mu (d_1 + d_2 \nabla^2) \frac{u}{r^2} Y_l^m$$
(15)

where we have put

$$p_0^2 = -2\mu w = 2\mu |w|. ag{16}$$

It is more convenient to work with the dimensionless variable

$$y = 2p_0 r (17)$$

and

$$\nabla_{y}^{2} = (1/4 \ p_{0}^{2}) \ \nabla^{2} \tag{18}$$

in terms of which eqn. (15) is given by

$$\left(1 + 4p_0^2 d_3 \nabla_y^2\right) \left(\frac{-u}{4y} - \frac{l(l+1)}{y^3} u + \frac{1}{y} \frac{d^2u}{dy^2}\right) Y_l^m
= \frac{\alpha \mu}{2p_0} (d_1 + 4p_0^2 d_2 \nabla_y^2) \frac{u}{y^2} Y_l^m.$$
(19)

To guess a suitable form for u, we observe that for a given $\alpha (= 2l + 1)$, the Laguerre polynomials $L_k^{\alpha}(y)$, where k (= n - l - 1) = 0, 1, 2,, form a complete set of orthogonal functions in y space $(0 < y < \infty)$ with respect to the weight functions e^{-y} $y^{(\alpha+l)}$. So u(y) can certainly be written as

$$u(y) = u_l(y) = N e^{-y/2} y^{(\alpha+1)/2} \sum_{n>l+1}^{\infty} b_n L_{n-l-1}^{\alpha}(y)$$
 (20)

where the coefficients b_n are as such arbitrary and N is just the usual normalisation constant. It is preferable to use, in place of b_n , the coefficients a_{n-1}/n^2 , where the denominator n^2 is 344 L.K. PANDE

chosen for later convenience and the choice of the index n-1 on a is useful because for a given l, the first coefficient in the series is simply a_l (to be followed by a_{l+1}, a_{l+2}, \ldots) and we can thus easily keep track of the particular angular momentum value we are dealing with. So we have

$$u_l(y) = Ne^{-y/2} y^{(\alpha+1)/2} \sum_{n\geq l+1}^{\infty} \frac{a_{n-1}}{n^2} L_{n-l-1}^{\alpha}(y)$$
 (20a)

This may also be written as

$$u_l(y) = Ne^{-y/2} y^{(\alpha+1)/2} \sum_{k=0}^{\infty} \frac{a_{k+l}}{(k+l+1)^2} L_k^{\alpha}(y)$$

$$=V_{l}\sum_{k=0}^{\infty}\hat{\mathcal{L}}_{k}^{\alpha}(y) \tag{20b}$$

where

$$\alpha = 2l + 1, \quad k = n - l - 1$$
 (21)

and we have put

$$\frac{a_{k+l}}{(k+l+1)^2} L_k^{\alpha}(y) = \hat{L}_k^{\alpha}, Ne^{-y/2} y^{(\alpha+1)/2} = V_l.$$
 (22)

On substituting the above expression for u in eqn. (19), and using the equation followed by the Laguerre polynomials, namely.

$$y \frac{d^{2} L_{k}^{\alpha}(y)}{dy^{2}} + (\alpha + 1 - y) \frac{dL_{k}^{\alpha}(y)}{dy} + kL_{k}^{\alpha}(y) = 0,$$
 (23)

we get
$$q \left(1 + 4 p_0^2 d_3 \nabla_y^2\right) \frac{V_l \hat{L}_k^{\alpha}}{y^2} Y_l^n = \frac{\lambda}{2p_0} \left(d_1 + 4 p_0^2 d_2 \nabla_y^2\right) \frac{V_l \hat{L}_k^{\alpha}}{y^2} Y_l^n \quad (24)$$

where
$$\lambda = -\alpha \mu$$
, $q = k + l + 1$. (25)

The sum over k in eqn. (24) has been suppressed for the present and will be put in later. It can easily be checked that

$$\nabla_{y}^{2} \frac{V_{l} \hat{L}_{k}^{\alpha}}{y^{2}} Y_{l}^{m} = \left[\frac{1}{4y^{2}} - \frac{(k+l)}{y^{3}} - \frac{2l}{y^{4}} \right] V_{l} \hat{L}_{k}^{\alpha} Y_{l}^{m} - \frac{2}{y^{3}} V_{l} \frac{d\hat{L}_{k}^{\alpha}}{dy} Y_{l}^{m}. \tag{26}$$

With the help of the standard recurrance relations

$$y \frac{dL_k^{\alpha}}{dy} = k L_k^{\alpha} - (k + 2l + 1) L_{k-1}^{\alpha}, \qquad (27)$$

$$y \frac{dL_k^{\alpha}}{dy} = (k+l) L_{k+l}^{\alpha} - (k+2l+2-y) L_k^{\alpha}, \qquad (28)$$

the last term in eqn. (26) can be written as

$$\frac{-2}{y^4} y \frac{d\hat{L}_k^{\alpha}}{dy} V_l Y_l^m = \left[\frac{2l+2}{y^4} \hat{L}_k^{\alpha} + \frac{(k+2l+1)}{y^4} \hat{L}_{k-1}^{\alpha} - \frac{(k+1)}{y^4} \hat{L}_{k+l}^{\alpha} - \frac{1}{y^3} \hat{L}_k^{\alpha} \right] V_l Y_l^m$$
 (29)

where

$$\hat{L}^{\alpha}_{k\pm i} = \frac{a_{k+l}}{(k+l+1)^2} L^{\alpha}_{k\pm i}; \quad i = 0 \pm 1, \pm 2, \dots$$
 (30)

So we get

$$\nabla_{y}^{2} \frac{V_{l} \hat{\mathcal{L}}_{k}^{\alpha}}{y^{2}} Y_{l}^{m} = \left[\frac{1}{4y^{2}} \hat{\mathcal{L}}_{k}^{\alpha} - \frac{(k+l+1)}{y^{3}} \hat{\mathcal{L}}_{k}^{\alpha} + \frac{2}{y^{4}} \hat{\mathcal{L}}_{k}^{\alpha} + \frac{(k+2l+1)}{y^{4}} \hat{\mathcal{L}}_{k-1}^{\alpha} - \frac{(k+1)}{y^{4}} \hat{\mathcal{L}}_{k+1}^{\alpha} \right] V_{l} Y_{l}^{m}$$

$$= \left[\frac{1}{4y^{2}} \hat{\mathcal{L}}_{k}^{\alpha} + A_{k,l} \right] V_{l} Y_{l}^{m} . \tag{31}$$

On substituting this result in eqn. (24), we get

$$(1 + p_0^2 d_3)q \frac{V_l \hat{L}_k^{\alpha}}{v^2} Y_l^m + 4p_0^2 d_3 q A_{k,l} V_l Y_l^m$$

346 L.K. PANDE

$$= \frac{\lambda}{2p_0} \left(d_1 + p_0^2 d_2 \right) \frac{V_l \hat{L}_k^{\alpha}}{y^2} Y_l^m + \frac{\lambda}{2p_0} 4p_0^2 d_2 A_{k,l} V_l Y_l^m. \tag{32}$$

On multiplying throughout by y^2 , dropping the common factor $V_l Y_l^m$, and using

$$C_1 = 1 - d_3 p_0^2$$
, $C_2 = (\lambda/2p_0) (d_1 - d_2 p_0^2)$

$$C_3 = 1 + d_3 p_0^2$$
, $C_4 = (\lambda/2p_0) (d_1 + d_2 p_0^2)$ (33)

so that $p_0^2 d_3 = (C_3 - C_1)/2$, $(\lambda/2 p_0) d_2 p_0^2 = (C_4 - C_2)/2$, (34)

eqn. (32) can be put in the form

$$(qC_3 - C_4) y^2 \hat{L}_k^{\alpha} + 2 [q(C_3 - C_1) - (C_4 - C_2)] [(q+l) \hat{L}_{k-1}^{\alpha} + 2\hat{L}_k^{\alpha} - (q-l) \hat{L}_{k+1}^{\alpha} - qy \hat{L}_k^{\alpha}].$$
(35)

The factors y and y^2 can now be removed by the application of the recurrance relation

$$yL_k^{\alpha}(y) = 2 q L_k^{\alpha}(y) - (q - l) L_{k+1}^{\alpha}(y) - (q + l) L_{k+1}^{\alpha}(y).$$
 (36)

Eqn. (35) then takes the form

$$A(q, l) \hat{L}_{k+2}^{\alpha} + B(q, l) \hat{L}_{k+1} + C(q, l) \hat{L}_{k}^{\alpha} + D(q, l) \hat{L}_{k-1}^{\alpha} + E(q, l) \hat{L}_{k-2}^{\alpha} = 0 (37)$$

where

$$A(q, l) = [qC_3 - C_4] [q - l] [q - l + 1]$$

$$B(q, l) = [qC_3 - C_4] [(-2q) (q - l) - 2 (q - l) (q + 1)]$$

$$+ 2 [q (C_3 - C_1) - (C_4 - C_2)] [q (q - l) - (q - l)]$$

$$C(q, l) = [q C_3 - C_4] [4q^2 + (q - l) (q + l + 1) + (q + l) (q - l - 1)]$$

$$+ 2 [q (C_3 - C_1) - (C_4 - C_2)] [-2q^2 + 2]$$
(38)

$$\begin{split} D\left(q,l\right) &= \left[q \ C_3 - C_4\right] \left[-2q \left(q+l\right) - 2 \left(q-1\right) \left(q+l\right)\right] \\ &+ 2 \left[q \left(C_3 - C_1\right) - \left(C_4 - C_2\right)\right] \left[q \left(q+l\right) + \left(q+l\right)\right] \\ E\left(q,l\right) &= \left[q C_3 - C_4\right] \left[q+l\right] \left[q+l-1\right]. \end{split}$$

Adding and substituting 2 (q - l), 4 and 2 (q + l), respectively, in the second brackets of the first terms of B(q, l), C(q, l) and D(q, l), and doing the cancellations that follow, we get,

$$B(q, l) = [qC_3 - C_4] [-2(q-l) (q+2)] - 2 [qC_1 - C_2] [(q-l) (q-1)]$$

$$= B_1 (q, l) + B_2 (q, l)$$

$$C(q, l) = [qC_3 - C_4] [2 (q-l) (q+l) - 2 (l-2)]$$

$$- 2 [qC_1 - C_2] [-2 (q+1) (q-1)]$$

$$= [q C_3 - C_4] [(q-l) (q+2) (q+l+1)]/q$$

$$+ [qC_3 - C_4] [(q-l-1) (q-2) (q+l)]/q$$

$$- 2 [qC_1 - C_2] [-2 (q+1) (q-1)]$$

$$= C_1(q, l) + C_2 (q, l) + C_3 (q, l)$$

$$D(q, l) = [qC_3 - C_4] [-2 (q+l) (q-2)] - 2 [q C_1 - C_2] [(q+l) (q+1)]$$

$$= D_1 (q, l) + D_2 (q, l)$$
(39)

where, in the breaking up of C(q, l), we have used

$$2(q-l)(q+l) - 2(l-2) = (q+l+1)(q+2)(q-l)/q + (q+l)(q-2)(q-l-1)/q.$$
 (40)

Eqn. (37) can thus be written as

$$\hat{L}_{k-2}^{\alpha}~E\left(q,l\right)~+~\hat{L}_{k-1}^{\alpha}~D_{2}\left(q,l\right)~+~\hat{L}_{k}^{\alpha}~C_{2}~\left(q,l\right)~+$$

348 L.K. PANDE

$$+ \hat{L}_{k-1}^{\alpha} D_{1}(q, l) + \hat{L}_{k}^{\alpha} C_{3}(q, l) + \hat{L}_{k+1}^{\alpha} B_{1}(q, l)$$

$$+ \hat{L}_{k}^{\alpha} C_{1}(q, l) + \hat{L}_{k+1}^{\alpha} B_{2}(q, l) + \hat{L}_{k+2}^{\alpha} A(q, l) = 0 .$$

$$(41)$$

Let us now put in the sum over k which had been suppressed so far for notational convenience. We also note that

$$q = k + l + 1$$

$$\hat{L}^{\alpha}_{k \pm i} = \frac{a_{k+l}}{(k+l+1)^2} \hat{L}^{\alpha}_{k \pm i}; \quad i = 0, \pm 1, \pm 2 \dots$$

Thus the last term in eqn. (41) can, for instance, be written as

$$\sum_{k=0}^{\infty} \hat{L}_{k+2}^{\alpha} A(q, l) = \sum_{k=0}^{\infty} \frac{a_{k+l}}{(k+l+1)^2} [qC_3 - C_4] [q-l] [q-l+1] L_{k+2}^{\alpha}$$

$$= \sum_{q=l+1}^{\infty} \frac{a_{q-1}}{q^2} [qC_3 - C_4] [q-l] [q-l+1] L_{q-l+1}^{\alpha}$$

$$= \sum_{q=l+3}^{\infty} [q=l-1] \frac{[q-2]}{[q-1]} \left[\left(C_3 - \frac{C_4}{[q-2]} \right) \frac{[q-l-2] [q-1]}{[q-2]^2} a_{q-3} \right] L_{q-l-1}^{\alpha} \quad (42)$$

Expressing all the sums in eqn. (41) in terms of L_{q-l-1}^{α} in a similar way, and remembering that $L_{q-l-1}^{\alpha} \neq 0$ only for $q \geq l+1$, we get

$$\sum_{q \geq l+1} \frac{[q+l+1][q+2]}{[q+1]} \left[\left[C_3 - \frac{C_4}{[q+2]} \right] \frac{[q+l+2][q+1]}{[q+2]^2} \ a_{q+1} \right. \\ \left. - 2 \left[c_1 - \frac{c_2}{[q+1]} \right] a_q + \left[c_3 - \frac{c_4}{q} \right] \frac{[q-l][q+1]}{q^2} \ a_{q-1} \right\} L_{q-l-1}^{\alpha} \\ \left. - 2 \sum_{q \geq l+1} \frac{[q+1][q-1]}{q} \left\{ \left[C_3 - \frac{C_4}{[q+1]} \right] \frac{[q+l+1]q}{[q+1]^2} \ a_q - 2 \left[C_1 - \frac{C_2}{q} \right] a_{q-1} \right.$$

$$+ \left[C_3 - \frac{C_4}{[q-1]} \right] \frac{[q-l-1]q}{[q-1]^2} \ a_{q-2} \right\} \ L_{q-l-1}^{\alpha}$$

$$+ \sum_{q \ge l+2} \frac{[q-l-1][q-2]}{[q-1]} \left\{ \left[C_3 - \frac{C_4}{q} \right] \frac{[q+l][q-1]}{[q]^2} \ a_{q-1} \right.$$

$$- 2 \left[c_1 - \frac{c_2}{[q-1]} \right] a_{q-2} + \left[c_3 - \frac{c_4}{[q-2]} \right] \frac{[q-l-2][q-1]}{[q-2]^2} a_{q-3} \right\} \ L_{q-l-1}^{\alpha} = 0. \tag{43}$$

It can be easily checked now that the solution of this equation is the simple difference equation

$$\left[c_{3} - \frac{c_{4}}{[q+1]}\right] \frac{[q+l+1]}{[q+l]^{2}} q a_{q} - 2\left[c_{1} - \frac{c_{2}}{q}\right] a_{q-1} + \left[c_{3} - \frac{c_{4}}{[q-1]}\right] \frac{[q-l-1] q}{[q-1]^{2}} a_{q-2} = 0;$$

$$q \ge l+1, \ a_{l-1} = 0$$

or, equivalently

$$\left[c_{3} - \frac{c_{4}}{[q+2]}\right] \frac{[q+l+2]|q+1|}{[q+2]^{2}} a_{q+1} - 2\left[c_{1} - \frac{c_{2}}{[q+1]}\right] a_{q} + \left[c_{3} - \frac{c_{4}}{q}\right] \frac{[q-l]|q+1|}{q^{2}} a_{q-1} = 0; (44b)$$

$$q \ge l, a_{l-1} = 0.$$

If we define a new set of coefficients \boldsymbol{b}_q by

$$b_q = a_q \frac{[q+l+1]}{[q+1]^3} [(q+1) c_3 - c_4], \qquad (45)$$

eqn. (44b) can be transformed to

350 L.K. PANDE

$$b_{q+1} - b_q \; \frac{2 \; [q+1]}{[q+l+1]} \frac{[(q+1) \; c_1 - c_2]}{[(q+1) \; c_3 - c_4]} \; + \; b_{q-1} \; \frac{[q-l]}{[q+l]} = 0; \; q \geq l, \; b_{l-1} = 0 \eqno(46)$$

his last form is convenient for computer calculations.

Eqn. (46) is of course identical to the difference equation which led us earlier to the sound state spectrum at high temperature¹. It is hoped that the elementary derivation of he equation presented here will be useful to anyone interested in exploring the bound state dea at finite temperature.

References

- 1. Malik, G.P., Pande, L.K. & Varma, V.S. (1991) Ap J. 379: 788.
- 2. Malik, G.P., Malik, U. & Varma, V.S. (1993) Ap & SS, 199: 299.
- 3. Malik, G.P., Malik, U., Pande, L.K. & Varma, V.S. (1995) Ap. J. 447: 443.
- 4. Pande, L.K. (1991) Pramana J. Phys. 37: 39.
- 5. Pande, L.K. Submitted for publication.
- Powell, J.L. & Crasemann, B. (1961) Quantum Mechanics, Reading, Massachusetts: Addison Wesley, Chapter 7.

Synthesis and antimicrobial activity of some new chalkones

Received July 14, 2000; Revised June 15, 2001; Accepted August 25, 2001

Abstract

2'-Hydroxy-1-(substituted phenyl)-3-(4-bromo-1-naphthyl)-2-propen-1-one were prepared by employing Claisen-Schmidt condensation. Structures of these compounds were confirmed by IR, PMR spectra and elemental analysis. All these compounds were found to be antibacterial and antifungal.

(Keywords: IR/PMR/elemental analysis/chalkones/antimicrobial agents)

Chalkones display interesting biological activities, including bactericidal¹⁻⁵, fungicidal⁶, antimalarial⁷, antiinflammatory⁸, cytotoxic⁹, anticancer¹⁰ and anti-invasive¹¹ properties. Hydroxychalkones have been widely investigated due to their synthetic potential and biological activity. In our continuing study of biologically active heterocyclic compounds like flavanones, flavones, flavonols, 2-pyrazoline and 1, 5-benzothiazepines, we discovered that the hydroxychalkones which serve as precursors of these heterocycles displayed potent antibacterial and antifungal activities against several human and plant pathogens. However, 2' or 4'-hydroxychalkones possessing naphthyl ring have not received considerable attention. Hence, we are reporting some new chalkones synthesised by a base-catalysed Claisen-Schimdt condensation of appropriately substituted 2' or 4'-hydroxyacetophenone and 4-bromo-1-naphthaldehyde.

2'-Hydroxy substituted acetophenone on treating with 4-bromo-1-naphthaldehyde in presence of ethanolic potassium hydroxide for 16h and on acidification with hydrochloric acid gave respective chalkone. They gave violet red colouration with ferric chloride and positive Wilson test¹². Infrared spectra of these compounds in nujol mull showed a band in the region 1620-1650 cm⁻¹ due to carbonyl group and a band between 1610-1630 cm⁻¹ due to ethylenic double bond. These assignments are in agreement with those observed by Mohanty *et al.*¹³, Dhar and Gupta¹⁴ and Vibhute and Jagdale¹⁵.

PMR spectra of a few representative chalkones were recorded in CDCl₃ on Varian T-60 spectrometer using TMS as an internal reference. Compound-2 (Table 1) showed singlet at $\delta 2.05$ due to Ar-CH₃ and multiplet at $\delta 6.9$ to $\delta 8.77$ due to aromatic and ethylenic protons and one singlet at $\delta 12.8$ due to a phenolic proton. These observations are in agreement with those observed by Cardillo *et al.*¹⁶

Table 1 - Analytical data and antimicrobial activity of chalkones,

Com No.	Ar C	Elemental analysis		Antibacterial activity		Antifungal activity			
		C, H & Obsd. (Calcd.)	Obsd.	Diameter of Zone of Inhibition in mm		Germination (%) after 12 h			
				E. coli	S. aureus	C. lunata		H. oryzae	
						% ger- mina- tion	length of germ tube (μ)	% ger- mina- tion	length of germ tube (µ)
1.	2'-OH .	C 64.76 (64.59) H 3.82 (3.68)	22.24 (22.66)	-	_	100	400	100	180
2.	2′-OH, 5′-CH₃	C 65.10 (65.40) H 3.85 (4.09)	21.67 (21.80)	16		100	380	100	190
3.	2'-OH, 5'-Cl		29.40 (29.81)	-	-	100	320	100	240
4.	2'-OH, 5'-Br		36.70 (37.04)	42	33	30	48	90	352
5.	2'-OH, 4'CH ₃ , 5'-Cl		29.25 (29.50)	18	10	95	310	80	280
6.	2'-OH, 3', 5'-di Cl		35.95 (35.78)	22	40	100	221	100	272
7.	2'-OH, 3', 5'-di I		55.10 (55.21)	33	30	20	32	100	144
8.	4'-OH		22.27 (22.66)	44	31	35	112	90	150
9.	4'-OH, 3',5'-di I Tetracycline Water-ethanol		55.45	30 48	26 25	50	80	20	96
	(90 : 10, v/v)					95	305	105	350

The antibacterial and antifungal activity of all synthesised compounds were assessed following the standard methods ¹⁷ against two animal pathogens *Staphylococcus aureus*, *Escherichia coli* and two fungal phyto pathogens e.g., *Curvularia lunata* and *Helminthosporium oryzae*, respectively. For antibacterial activity, the filter paper discs were soaked in the solution of above compounds at a concentration of 150 ppm of compound in 10% ethanol and placed at the centre of the bacteria seeded agar plates (petridishes).

The petridishes were incubated at 26±1°C for 24h. The strength is reported in mm (Table 1) and results were standardised against tetracycline.

For antifungal activity, spore suspension was prepared from 5 day old PDA (Potato dextrose agar) slope cultures. Spore suspensions were placed in small petridishes. Solutions of different synthesised compounds were prepared in 90:10 (v/v) water: ethanol, and the concentrations of compounds (150 ppm) were adjusted in spore suspensions. Petridishes were incubated for 12 h under moist chambers. Aqueous ethanol (90:10, v/v) served as control. Percentage germination and germ tube length with effect of these compounds after a period of 12h was recorded by observing petridishes directly under microscope. These results are given in Table 1.

References

- 1. Schrauufstatter, E. & Deutsch, S. (1948) Z. Natur forsch. 48:430.
- 2. Misra, S.S. (1971) Indian J. Appl. Chem. 34: 260.
- 3. Misra, S.S. (1973) J. Indian Chem. Soc. 50: 68.
- 4. Vibhute, Y.B. & Wadje, S.S. (1976) Indian J. Exp. Biol. 14: 739.
- 5. Vibhute, Y.B. (1976) J. Indian Chem. Soc. 53:736.
- 6. Buu-Hui, N.P. & Xuong, N.D. (1958) J. Org. Chem. 23: 39.
- Li, R., Kenyon, G.L., Cohen, F.E., Chen, X., Gong, B., Dominguez, J.N., Davidson, E., Kurzban, G., Miller, R.E. & Nuzman, E.O. (1995) J. Med. Chem. 38: 5031.
- Ballesteros, J.F., Sanz, M.J., Ubeda, A., Miranda, M.A., Iborra, S., Paya, M. & Alcaraz, M.J. (1995) J. Med. Chem. 38: 2794.
- 9. Yit, C.C. & Das, N.P. (1994) Cancer Lett. 82:65.
- 10. Dinkova-Kostova, A.T., Abeygunawardana, C. & Talalay, P. (1998) J. Med. Chem. 41: 5287.
- 11. Parmar, et al. (1998) Indian J. Chem. 37B: 628.
- 12. Wilson (1938) J. Amer. Chem. Soc. 61: 2303.
- 13. Mohanty, P., Pande, P., Sabata, B.K. & Raut, M.K. (1965) Indian J. Chem. 3: 121.
- 14. Dhar, D.N. & Gupta, V.P. (1971) J. Indian Chem. Soc. 9:818.

- 15. Vibhute, Y.B. & Jagdale, M.H. (1986) Acta ciencia India 12: 223.
- 16. Cardillo, G., Merlini, L. & Mondelli (1968) Tetrahedron 24:497.
- 17. Collins, C.H. (1967) Microbiological methods, Butterwords, London p. 364.

G.R. SUBBANWAD, M.A. BASEER and Y.B. VIBHUTE*

*P.G. Department of Chemistry, Yeshwant College, Nanded-431 602, India.

Vol. 71, Part I, 2001

CONTENTS

Chemistry		
Equalized electronegativity and chelation		
B.S. Sekhon	•••	1
Studies on synthesis of some novel bifunctional reactive dyes and their application on various fibres		
M.S. Patel, S.K. Patel and K.C. Patel	•••	5
Study of metal drug equilibria in solution		
Purushottam B. Chakrawarti, Mukta Chakrawarti and Pramila Maini	•••	13
Mathematics		
Thermo-elastic stress distribution in three layered system		
Ashoke Das and Bimalendu Das		21
Effect of rotation on thermal instability in Walters elastico-viscous fluid		
Pardeep Kumar	•••	33
Physics		
Possibility of the simultaneous occurrence of potential minima along two crystallagrophic directions in an octahedral potential		
R.K. Tiwari and Prem N. Singh	•••	43
Liquid crystalline behaviour of alkoxybenzoic acids: A theoretical study		
S.N. Tiwari and Nitish K. Sanyal	•••	53
Dielectric study of pyridine-alcohol binary liquids at 25°C		
Ajay Chaudhari, Sunil Ahire, Milind Lokhande and Suresh Mehrotra		75
and bureon internotia	•••	, ,

Vol. 71, Part II,2001

Chemistry		
Synthesis and characterization of transition metal complexes of isatin oxime		
K. Laxmi, B. Sireesha and Ch. Sarala Devi	•••	85
Synergic extraction and trace determination of vanadium (V) as ternary chelate with cinnamohydroxamic acid and thiocyanate in mibk and its application		
A.K. Chakrabarti		95
Synthesis and characterization of silatranes containing heterocyclic moieties: Part III		
Shailesh C. Verma, M. Nasim and P.S. Venkataramani	•••	105
Mathematics		
Stability of a stratified partially ionized plasma		
Aiyub Khan and P.K. Bhatia	•••	113
Boundary value problems of infinite plate weakened by a curvilinear hole Abdullah A. Badr	•••	125
Motions and affine motions in Finsler, Randers and Kropina Spaces U.P. Singh	•••	135
Free convection in MHD flow of a rotating viscous liquid in porous medium past a vertical porous plate		
N.P. Singh, Sarvesh Kumar Gupta and Atul Kumar Singh		149
On the ultradistribution of Mellin transform		
P.K. Banerji and Deshna Loonker		159
Short Notes		
Mixed ligand complex formation of M ²⁺ ions and quinoline and thymine Purushottam B. Chakrawarti and Ms. Sandhya Sharma	•••	173
A new spot test for the on field detection of dithiocarbamate fungicides H.S. Rathore, S. Mital, S. Kumar and Y.N. Singh		177
Study of metal-drug equilibria in solution		
Purushottam B. Chakrawarti and Km. Sandhya Sharma		185

Vol. 71, Part III, 2001

Chemistry		
Synthesis of some new azo-coumarins and Schiffs bases as possible antibacterial agents		
Sangeeta Bhatnagar, Kausik Ghosh and D.S. Seth	•••	189
On the coordinating behaviour of sulphamethoxazole towards some metal ions		
B.S. Sekhon, Harvinder K. Sahai and H.S. Randhawa		195
Synthesis and characterization of 8-hydroxy-quinoline-oxamide- formaldehyde tercopolymers		
Peter S. Lingala, L.J. Paliwal and H.D. Juneja	•••	205
Mathematics		
A uniform model for the storage utilization of B-tree-like structure P.K. Mishra and C.K. Sharma	•••	213
Non-linear effects in the motion and stability of an inter-connected satellites system orbiting an oblate Earth		
B.M. Singh, Ashutosh Narayan and R.B. Singh		225
A common fixed point theorem in complete metric spaces by altering distances		
K.P.R. Sastry and G.V.R. Babu		237
Hall current effects on heat and mass transfer in flow of a viscous fluid N.P. Singh		243
Unsteady flow of visco-elastic Oldroyd fluid with transient pressure gradient through a rectangular channel		
Shyamal Kumar Kundu and P.R. Sengupta	•••	253
Fluctuating free convective flow with radiation through porous medium having variable permeability		
S.S. Tak and Arvind Maharshi		263
Physics		
Study of photoconductivity in (B ₂ O ₃ – ZnS) Cu (0.1%) Cl (1%) composite Akhilesh Kumar, Sunil Kumar Srivastava and S.G. Prakash		273
Short Note		
Study of drinking water quality in Gorakhpur city		
Alok Srivastava, Amitabh K. Srivastava, P.P. Srivastava and S.C. Prasad		281

Author Index

Ahire, Sunil	75	Patel, K.C.	5
Babu, G.V.R.	237	Patel, M.S.	5
Badr, Abdullah A.	125	Patel, S.K.	5
Banerji, P.K.	159,321	Prakash, S.G.	273
Baseer, M.A.	351	Prasad, S.C.	281
Bhatia, P.K.	113	Randhawa, H.S.	195
Bhatnagar, Sangeeta	189	Rathore, H.S.	177
Chakrabarti, A.K.	95	Sahai, Harvinder K.	195
Chakrawarti, Mukta	13	Sanyal, G.S.	295
Chakrawarti, Purushottam B	.13,173,185	Sanyal, Nitish K.	53
Chaudhari, Ajay	75	Sastry, K.P.R.	237
Das, Ashoke	21	Sekhon, B.S.	1,195
Das, Bimalendu	21	Sengupta, P.R.	253
Devi, Ch. Sarala	85	Seth, D.S.	189
Ganguly, R.	295	Sharma, C.K.	213
Ghosh, Kausik	189	Sharma, Km. Sandhya	173,185
Gupta, Sarvesh Kumar	149	Shenan, Gamal M.	321
Juneja, H.D.	205	Singh, Atul Kumar	149
Khan, Aiyub	113	Singh, B.M.	225
Khurana, Sandhya	305	Singh, N.P.	149,243
Kumar, Akhilesh	273	Singh, Prem N.	43
Kumar, Pardeep	33	Singh, R.B.	225
Kumar, S.	177	Singh, Sarva Jit	305
Kundu, Shyamal Kumar	253	Singh, U.P.	135
Laxmi, K.	85	Singh, Y.N.	177
Lingala, Peter S.	205	Sireesha, B.	85
Lokhande, Milind	75	Sivaramakrishna, Akella	289
Loonker, Deshna	159	Srivastava, Alok	281
Maharshi, Arvind	263	Srivastava, Amitabh K.	281
Maini, Pramila	13	Srivastava, P.P.	281
Mehrotra, Suresh	75	Srivastava, Sunil Kmar	273
Mishra, P.K.	213	Subbanwad, G.R.	351
Mital, S.	177	Tak, S.S.	263
Narayan, Ashutosh	225	Tiwari, R.K.	43
Nasim, M.	105	Tiwari, S.N.	53
Nath, P.K.	295	Venkataramani, P.S.	105
Paliwal, L.J.	205	Verma, Shailesh C.	105
Pande, L.K.	339	Vidhute, Y.B.	351

Subject Index

4,4'-methylene-bis ortho		Hall current	243
anisidine	5	Hane-transform	21
8 HQQF	205	Heat flux	21
Alcohols	75	Heat source	243
Alkoxybenzoic acids	53	Heterocyclic substituted silatra	anes 105
Antimicrobial agents	351	Hilbert space	237
Axisymmetric	21	Hydromagnetic stability	113
Bis-azo reactive dyes	5	IR	5,351
B-trees	213	Inter connected satellities	225
CNDO	53	lonic crystals	43
Cation-exchange resin	177	Isatin oxime	85
Cd(II)	13	Kropina space	135
Cephalexin complexes	13	Layered problem	21
Chalkones	351	Linear temperature distribution	1 21
Chelation	1	Liquid crystals	53
Chemistry 1,5,13,85,9	5.105.	M2+-quinoline-tymine equilibri	a 173
173,177,185,18		Magnetic field	149
205,281,289,29		Magnetic & spectral studies	85
Common fixed points	237	Mass transfer	243
Complete matric space	237	Mathematics 21,33,113,1	125,135,
Composites	273	149,159,213,2	225,237,
Convex and starlike convex		243,253,263,	305,321
functions	321	Mellin transform	159
Coordinating behaviour	195	Mesogen	53
Copper-ammonia complex	177	Metal complexes	295
Coumarin derivatives	189	Metal ions	195
Dielectric parameters	75	Metal-drug equilibria	13,185
Dithiocarbamates	177	Mg(II) complexes	13
Earth oblateness	225	Mixed-ligand complexes	13
Elasto-static	21	Molecular interactions	53
Electron proton pairing	339	Monoclinic media	305
Electronegativity	1	Non-linear oscillation	225
Elemental analysis	351	N-CHA	95
Ferricyanide	289	N-(2-tri fluoromethyl)	189
Finite resistivity	113	Octahedral potential	43
Finite temperature theory	339	PMR	351
Finsler geometry	135	Phenyl malonamic acid	189
Fractional differential and		Photoconductivity	273
integral operators	321	Physico-chemical &	
Fungicides	177	bacteriological properties	281
Generalized visco-elastic fluid	253	Physics 43,53,75,	273,339

Polarization	43	Tercopolymer	205
Porous medium	149	Ternary chelate	95
Pyrazine carboxylic acid		Ternary complexes 1	73,185
hydrazides	295	Testing function space	159
Pyridine	75	Thermal instability	33
Randers space	135	Time domain reflectometry	75
Rectangular channel	253	Titrimetry	289
Reflection	305	Transesterification	105
Riordan array	213	Transient pressure gradient	253
Rotating viscous liquid	149	Transition metal complexes	85
SCN	95	Transmission	305
Schiffs bases	189,295	UO2(VI)	295
Selfmaps	237	Ultradistribution	159
Semi-conductors	273	Uniform rotation	33
Sherwood number	243	Univalent and p-valent functions	321
Silk	5	Unsteady flow	253
Soret number	243	VO(IV)	295
Spectral studies	189	Vanadium	95
Stability constant	13	Viscose rayon	5
Stability of bivalent metal		Viscosity	113
complexes	173	Visco-elastic Oldroyd fluid	253
Stability of metal chelates	1	Walters elastico-viscous fluid	
Stability of quinoline complexe	es 173	(model B')	33
Stratified plasma	113	Water quality	281
Sulphamethoxazole	195	Waves	305
Sulphide sulphur	289	Wool	5
Synergic extraction	95	X-ray astronomy	339

.

EDITORIAL BOARD

Chief Editor Prof. H.C. Khare

Former, Professor of Mathematics, University of Allahabad; The National Academy of Sciences, India, 5, Lajpatrai Road, Allahabad – 211 002

Fax: 091-532-641183 E-Mail: nasi@nde.vsnl.net.in

- Prof. R.P. Agrawal
 Former, Vice-Chancellor,
 Rajasthan & Lucknow Universities,
 B1/201, Nirala Nagar,
 Lucknow 226 020
 (Mathematics)
- 3. Prof. Alok K. Gupta
 Head, Department of Earth and
 Planetary Sciences,
 University of Allahabad,
 Allahabad 211 002
 Fax: 091-532-641840
 E-Mail: ncemp@nde.vsnl.net.in
 (Earth Sciences)
- Prof. B.L. Khandelwal Emeritus Scientist (CSIR), Defence Materials and Stores Research and Development Establishment, DMSRDE Post Office, G.T. Road, Kanpur – 208 013

Fax: 091-512-450404 (Chemistry)

7. Prof. K.S. Valdiya
Bhatnagar Research Professor,
Jawaharlal Nehru Centre for
Advanced Scientific Research,
Jakkur P.O.
Bangalore – 560 064

Fax: 091-80-6462766; E-Mail: uday@jncasr ac.in

(Geology)

 Prof. Suresh Chandra Department of Physics, Banaras Hindu University, Varanasi – 221 005 Fax: 091-542-317074

E-Mail: schandra@banaras.ernet.in

(Physics)

4. Prof. S.V. Kessar Senior Scientist, Department of Chemistry, Panjab University, Chandigarh – 160 014

E-Mail: sykessar@panjabuniv.chd.nic.in

(Chemistry)

Prof. U.C. Mohanty
 Professor & Head,
 Centre for Atmospheric Science,
 Indian Institute of Technology,
 Hauz Khas,
 New Delhi – 110 016

Fax: 091-11-6862037

E-Mail: mohanty@cas.iitd.ernet.in (Atmospheric Sciences)

Managing Editor Prof. S.L. Srivastava

Formerly Professor & Head, Department of Physics, University of Allahabad;
The National Academy of Sciences, India,
5, Laipatrai Road, Allahabad – 211 002

Fax: 091-532-641183 E-Mail: nasi@nde.vsnl net in

EDITORIAL ADVISORY BOARD

1 Prof. Edwin D. Becker Scientist Emeritus, National Institutes of Health, 5 Center Drive, Bethesda. Maryland 20892 - 0520, U.S.A.

E-Mail:tbecker@nih.gov

(Spectroscopy/Nuclear Magnetic Resonance)

3. Prof. S. Chandrasekhar Director. Centre for Liquid Crystal Research, P.B. No.1329, Bangalore - 560 013 Fax . 091-80-8382044 E-Mail:ucler@giasbg01.vsnl.net.in (Liquid Crystals/Condensed Matter)

5. Prof. M.G.K. Menon President, ISI & Past President ICSU, Chairman, Board of Governors, IIT (Bombay), C-63, Tarang Apts., 19-I.P. Ext., Mother Dairy Road, Patpargani,

Delhi - 110 092 Fax: 091-11-6959456 E-Mail: mgkmenon@ren02.nic in

(Physics)

7. Prof. C.Kumar N. Patel Chairman & CEO. Pranalytica Inc., 1101 Colorado Avenue. Santa Monica, CA90401 - 3009, U.S.A., Fax: 310-458-0171 (Physics)

9. Dr. P. Rama Rao Vice-Chancellor, University of Hyderabad, P.O. Central University, Hyderabad - 500 046 (Physical and Mechanical Metallurgy)

11. Prof. Govind Swarup Homi Bhabha Senior Fellow. National Centre for Radio Astrophysics. Tata Institute of Fundamental Research. Pune University Campus, P. B. No. 3, Ganeshkhind, Pune - 411 007

Fax: 091-20-5657257, 5655149 E-Mail: gswarup@ncra tifr res in

(Radio Astronomy)

2. Prof. Sır Harmann Bondı 60 Mill Lane, Impington, Cambridge CB4 9XN, U.K. Tele: 01223 565180 (Mathematical Astronomy)

4. Prof. S.K. Joshi Sarabhai Research Professor, JNCASR, National Physical Laboratory, Dr. K.S. Krishnan Marg. New Delhi - 110 012 Fax: 091-11-5852678 E-Mail: skioshi@csnpl.ren.nic.in (Solid State Physics)

6. Prof. A.P. Mitra Formerly Secretary, DSIR and Director-General, CSIR, Hon, Scientist of Eminence, National Physical Laboratory, Dr. K.S. Krishnan Marg, New Delhi - 110 012 Fax: 091-11-5852678, 5764189

E-Mail: apmitra@doe.ernet.in; apmitra@ndf.vsnl.net.in (Ionospheric Physics/Radio Communication/ Space Physics/Space Science)

8. Dr. B.L.S. Prakasa Rao Distinguished Scientist, Indian Statistical Institute, 7, S.J.S. Sansanwal Marg. New Delhi - 110 016 Fax: 091-11-6856779 E-Mail: blsp@isid.ac.in (Mathematical Statistics)

10. Prof. M.M. Sharma Kothari Research Professor (Hon.), JNCASR, Formerly Director & Professor of Chemical Engineering, UDCT, 502, Saurabh, Plot No. 39, Kunder Marg, Swastik Part, Chembur, Mumbai - 400 071 E-Mail: mmsharma@bom.3 vsnl net in

12. Prof. H.C. Khare (Chief Editor) Former, Professor of Mathematics, University of Allahabad. The National Academy of Sciences, India. 5, Lapatrai Road, Allahabad - 211 002

Fax: 091-532-641183 E-Mail: nasi@nde.vsnl net.in

(Applied Mathematics/Theoretical Physics)